ZnO Materials as Effective Anodes for the Photoelectrochemical Regeneration of Enzymatically Active NAD+

1. Introduction

vs.

2. Experimental Section

2.1 Materials

Candida boidinii

2.2 Synthesis of ZnO nanowires (ZnOW)

2.3 Synthesis of porous ZnO (ZnOP)-based films

2.7 Regeneration of NAD^+

vs.

$$NADH \rightarrow NAD^+ + H^+ + 2\,e^-$$

$$Formate + NAD^+ \overset{FDH}{\longleftrightarrow} CO_2 + NADH + H^+$$

3. Results and discussion

3.1 Morphological and structural characterization of ZnO nanostructures and films

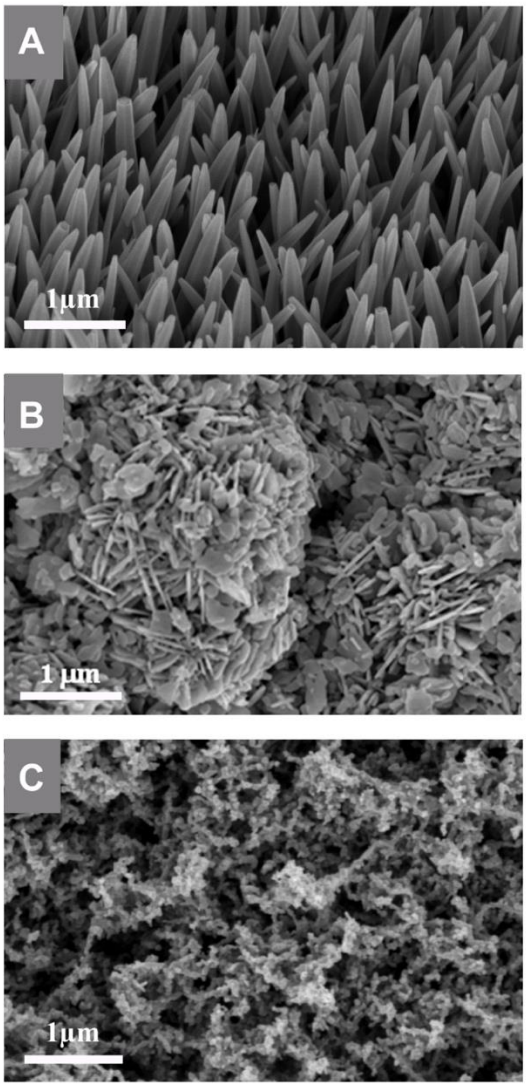

Figure 1.

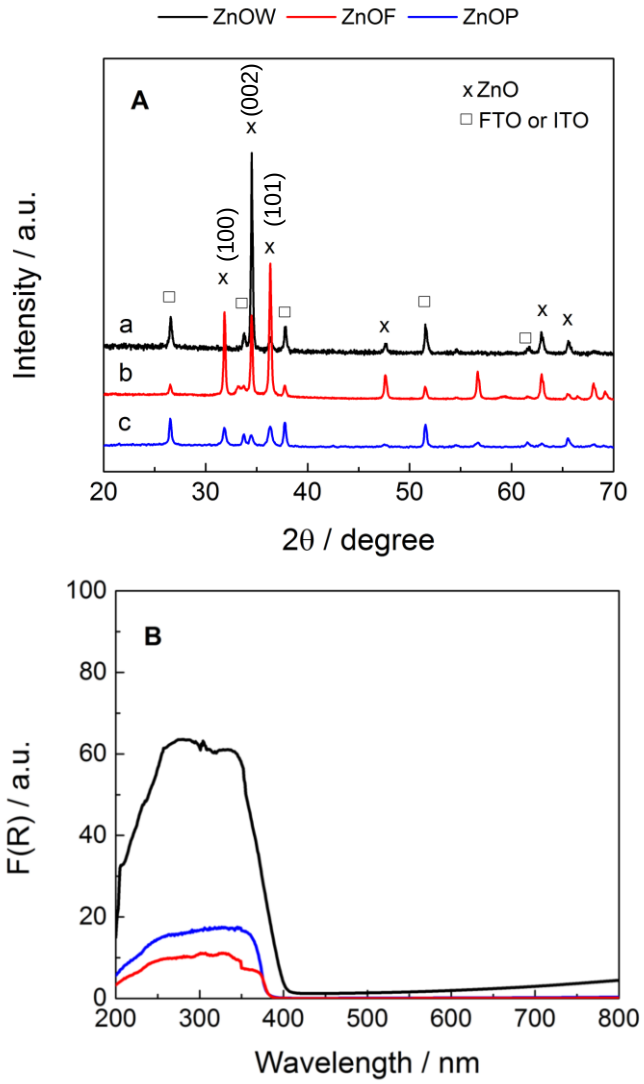
ZnOW
ZnOF
ZnOP
A
x ZnO
□ FTO or ITO
(002)
(100)
(101)
a
b
c
Intensity / a.u.
2θ / degree
B
F(R) / a.u.
Wavelength / nm

Figure 2.

Table 1.

			□□□□□□□□□□	

3.2. Electrochemical oxidation of NADH

vs.

vs.

vs.

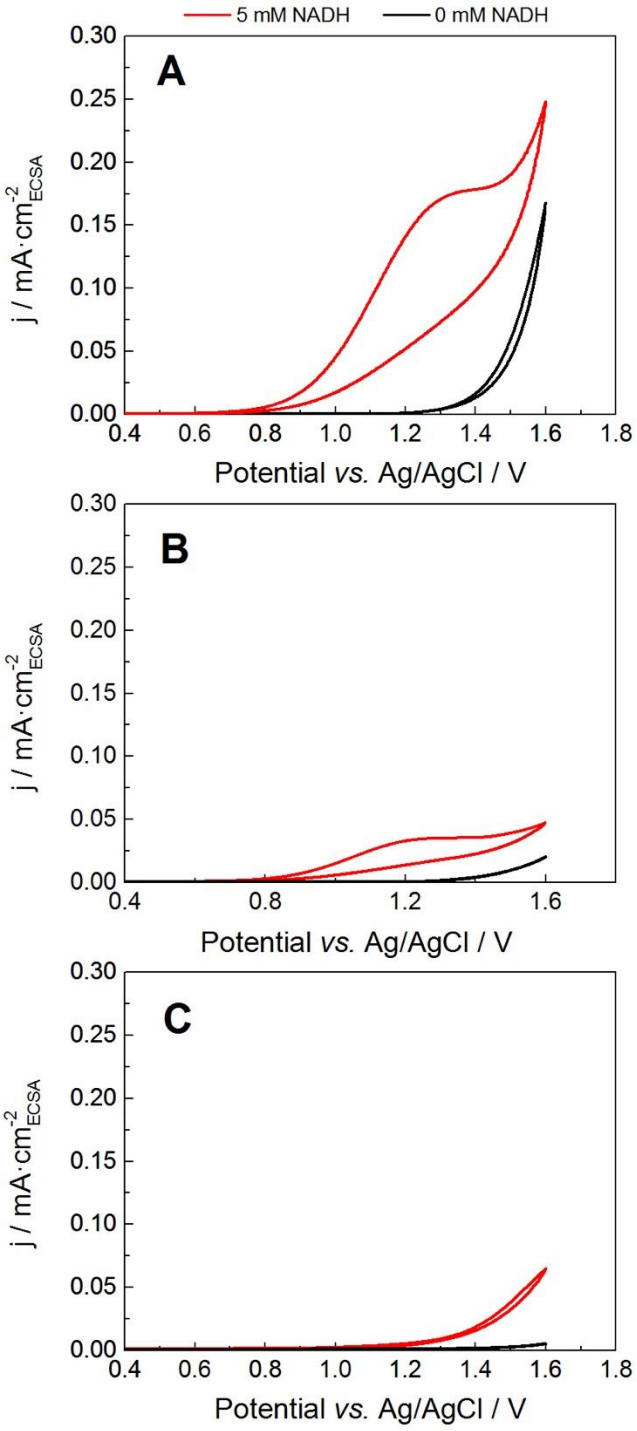
5 mM NADH
0 mM NADH
A
B
C
j / mA·cm-2ECSA
Potential vs. Ag/AgCl / V
0.00
0.05
0.10
0.15
0.20
0.25
0.30
0.4
0.6
0.8
1.0
1.2
1.4
1.6
1.8

Figure 3.

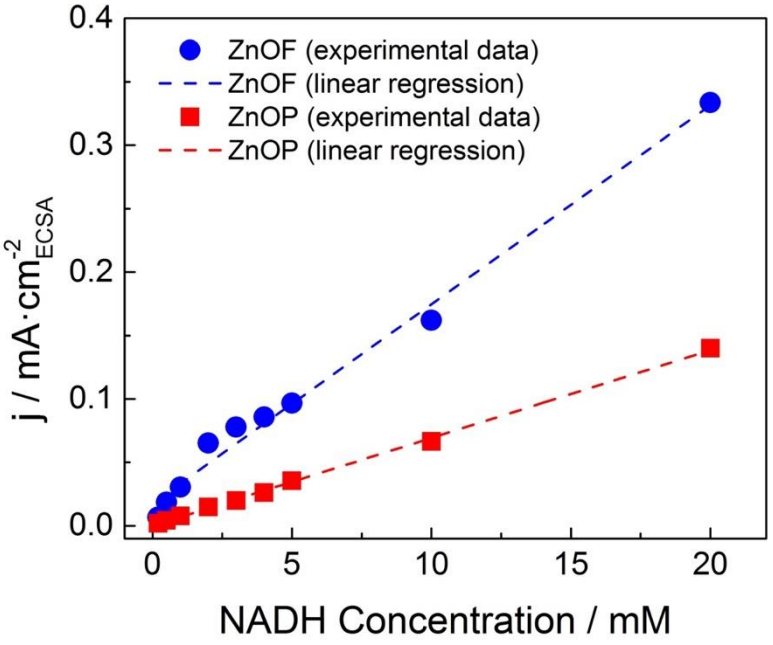

Figure 4.

3.3 Effect of the illumination

vs.

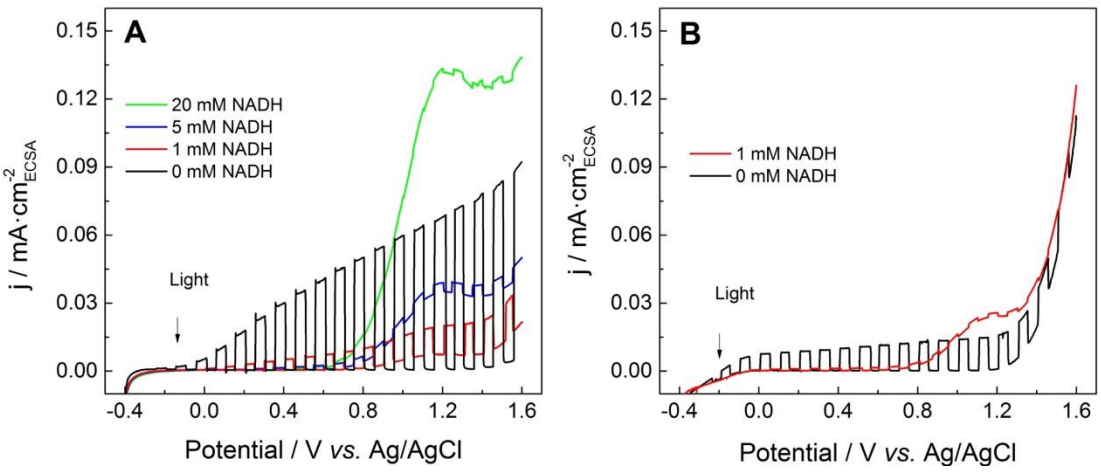
A
B
20 mM NADH
5 mM NADH
1 mM NADH
0 mM NADH
Light
j / mA·cm-2ECSA
Potential / V vs. Ag/AgCl
0.15
0.12
0.09
0.06
0.03
0.00
-0.4
0.0
0.4
0.8
1.2
1.6

Figure 5.

Table 2. □□□□□□□□□□

3.4 Photoelectrochemical oxidation of NADH

vs.

□□□□□□□ □□□□□□□

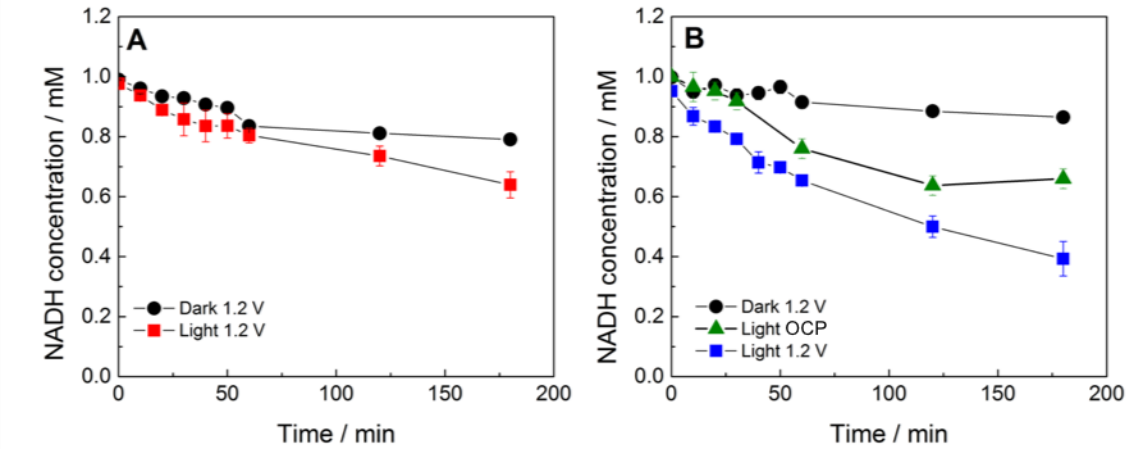

Figure 6.

vs.

vs.

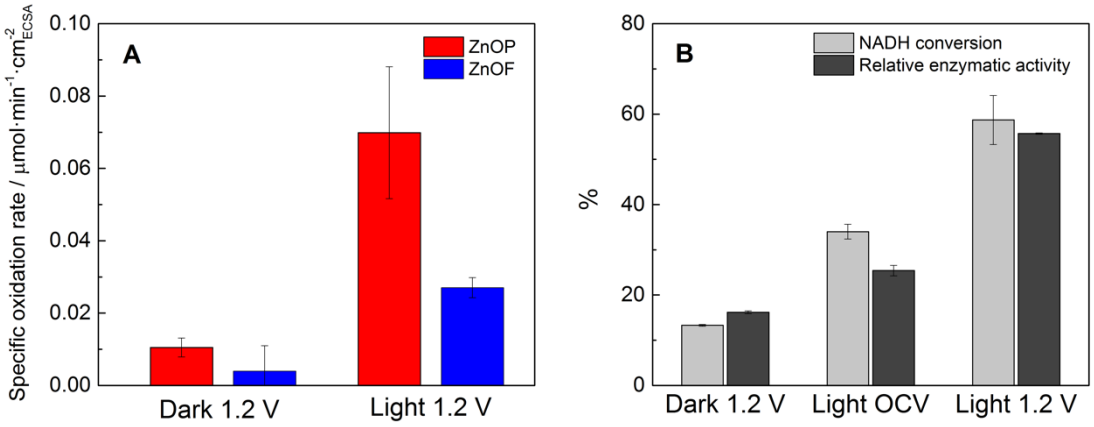
A
ZnOP
ZnOF
Specific oxidation rate / μmol·min-1·cm-2ECSA
0.10
0.08
0.06
0.04
0.02
0.00
Dark 1.2 V
Light 1.2 V
B
NADH conversion
Relative enzymatic activity
%
80
60
40
20
0
Dark 1.2 V
Light OCV
Light 1.2 V

Figure 7. *vs.*

3.5 Enzymatic activity of regenerated NAD

vs. *vs*.

vs.

4. Conclusion

Supporting Information ECSA calculation; Optical characterization of the electrodes.

Corresponding Author

Present Addresses

Author Contributions

Funding Sources

S. cerevisiae *J. Appl.*

Microbiol. **2018** *102* í

-

AIChE J

2018 *64* í

Glutamine and Glutamate

Mammals **2018** í

Appl. Microbiol. Biotechnol. **2017** *101* í

Candida boidinii

J. CO_2 Util. **2018** *28* í

Bioelectrochemistry **2004** *64* í

J. Mol. Catal. B: Enzym. **2014** *103* í

Joule **2020** *4* í

Biotechnol. Adv. **2007** *25* í

J. Mol. Catal. B: Enzym. **2015** *120* í

J. Mol. Catal., B Enzym. **2016** *134* í

Trends Biotechnol. **2006** *24* í

Int. J.

Electrochem. Sci. **2012** *7* í

Bioelectrochemistry **2002** *56* í

Biocatal. Biotransformation **2004** *22* í

□

J. Am. Chem. Soc. **2014** *136* í

Sens. Actuators, B **2011** *156* í

- *p-* *ChemElectroChem* **2017** *4*

í

Thermus thermophilus *BMC Biotechnol.*

2011 *11*

Protein Eng. Des.

Sel. **2018** *31* í

Phosphorus Recovery and Recycling **2019**

í

Appl. Microbiol.

Biotechnol. **2013** *97* í

In situ

J. Electroanal. Chem.

2003 *541* í

J.

Electrochem. Soc. **2015** *162* í

J. Appl. Electrochem. **2001** *31*

í

Anal. Chim. Acta. **2003** *476* í

Bioelectrochemistry **2000** *51* í

Bioresour. Technol. **2002** *82* í

Green Chem. **2013** *15* í

Encyclopedia of Electrochemistry **2007** í

Electrochem. Commun. **2011** *13* í

□□□□□□□□□ □□□□□□□□□ □□

Chem. Eng. Sci. **2017** *158* í

Nanowires: Synthesis,

Electrical Properties and Uses in Biological Systems **2014** í

Phys. Chem. Chem. Phys. **2015** *17*

í

Appl. Catal., B **2019** *243*

í

Energy Environ. Sci. **2008** *1* í

Nanomaterials **2017** *7*

Adv. Mater. Interfaces **2017** *4*

Appl. Phys.

A **2013** *113* í

New J. Chem. **2014** *38* í

Coatings **2018** *8*

Phys. Chem. Chem. Phys. **2014** *16* í

CrystEngComm **2016** *18*

í

Anal. Lett. **2008** *41* í

J. Nanopart. Res. **2014** *16*

Mater. Lett. **2009** *63*

í

Mater. Sci. Eng., C. **2015** *46*

í

Chem. Eur. J. **2013** *19* í

RSC Adv. **2016** *6* í

ACS Appl. Mater. Interfaces **2013** *5* í

J. Appl. Crystallogr. **1978** *11* □

Energy **2014** *65* í

Materials **2016** *9*

Materials

2018

Catalysts **2017** *7*

ACS Appl.

Mater. Interfaces **2014** *6* í

J. Appl. Electrochem.

2009 *39* í

For Table of Contents Only