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Human errors and measurement uncertainty

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## Human errors and measurement uncertainty in chemical analysis

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## Human errors and measurement uncertainty in chemical analysis

Ilya Kuselman <sup>a,\*</sup> and Francesca Pennechi <sup>b</sup>

<sup>a</sup> *National Physical Laboratory of Israel (INPL), Danciger "A" Bldg, Givat Ram, 91904  
Jerusalem, Israel*

<sup>b</sup> *Istituto Nazionale di Ricerca Metrologica (INRIM), 91 Strada delle Cacce, 10135  
Turin, Italy*

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\* Corresponding author. Tel.: +972-50-6240466; fax: +972-2-6303516.

*E-mail address:* [ilya.kuselman@gmail.com](mailto:ilya.kuselman@gmail.com) (I. Kuselman).

## Abstract

Evaluating the residual risk of human errors in chemical analysis, remaining after the error reduction by a laboratory quality system, and quantifying the consequences of this risk for the quality of chemical analytical results are discussed based on expert judgments and Monte Carlo simulations. A procedure for evaluation of the contribution of the residual risk to the measurement uncertainty budget is proposed. Examples are provided using earlier published sets of expert judgments on human errors in pH measurement of groundwater, elemental analysis of geological samples by inductively coupled plasma mass spectrometry, and multi-residue analysis of pesticides in fruits and vegetables. The human error contribution to the measurement uncertainty budget in the examples was not negligible, yet also not dominant. This was assessed as a good risk management result.

Keywords: chemical analysis, human error, expert judgment, measurement uncertainty, quality risk management

## 1. Introduction

### *1.1. Measurement error and human error*

The international vocabulary of metrology [1] defines measurement error as a difference between measured and reference quantity values, and stipulates that measurement errors should not be confused with mistakes. Since a mistake is a kind of human error, the meaning of this stipulation is the distinction of measurement error from human error.

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3 Human error in a chemical analytical (testing) laboratory is any action or lack thereof  
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5 that leads to exceeding the tolerances of the conditions required for the normative work  
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7 of the analytical measuring system with which the human interacts [2]. Such tolerances  
8  
9 (interval of temperature values for a sample decomposition, pH values for an analyte  
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11 extraction, etc.) are formulated in standard operation procedures of the analysis based  
12  
13 on results of the analytical method validation study.  
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16 Just as 20-30 years ago managers of chemical analytical laboratories were afraid that  
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18 evaluating and reporting measurement uncertainty [3-5] could compromise the  
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20 laboratory reputation, today any discussion of human errors in a laboratory is sensitive  
21  
22 also. One can even find an opinion that human errors in an analytical laboratory “are not  
23  
24 interesting for science and have no influence on uncertainty” [6]. However, there are a  
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26 number of factors effecting chemical analytical (measurement/test) results to various  
27  
28 degrees and human errors are a part of them. Gross errors are easily identifiable, and  
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30 corresponding results can be separated from the data set. At the same time, small human  
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32 errors are in principle not distinguishable from other components of measurement  
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34 uncertainty. Therefore, an uncertainty budget is not complete when consequences of  
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36 possible human errors of a sampling inspector and/or an analyst/operator are not taken  
37  
38 into account as a contribution to the budget [2, 7-9]. Moreover, consideration of human  
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40 errors is required now by national and international documents for correct evaluation of  
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42 quality of chemical analytical results in medicine, food and drug analysis and other  
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44 fields [10-14].  
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## 52 *1.2. Background*

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3 Risk of human error can be defined as a combination of the likelihood (probability) of  
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5 occurrence of the error in a chemical analysis and the severity of that error for the  
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7 quality of the analytical results [15]. A particular kind of human error and a certain step  
8  
9 of the analysis at which this error may happen, are considered as the error scenario.  
10  
11 Evaluation of likelihood and severity of an error scenario is possible on the basis of  
12  
13 expert judgments [16]. An expert in the analytical method can judge the likelihood of  
14  
15 error scenarios  $i = 1, 2, \dots, I$  by the following scale: likelihood of an unfeasible scenario  
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17 – as  $p_i = 0$ , weak likelihood - as  $p_i = 1$ , medium – as  $p_i = 3$ , and strong (maximal)  
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19 likelihood – as  $p_i = 9$ . A discussion of this scale is available in ref. [16]. Other scales  
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21 can also be used depending on the tasks. The normalized and averaged value of the  
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23 expert judgments  $P^* = (100\%/9) \sum_{i=1}^I p_i / I$  is a kind of “intuitive” or “subjective”  
24  
25 (mean) error probability [4]. The similarly calculated severity score  
26  
27  $L^* = (100\%/9) \sum_{i=1}^I l_i / I$ , where  $l_i$  is the expert judgment on the severity of error  
28  
29 scenario  $i$ , given again on the scale (0, 1, 3, 9), reflects the (mean) loss of quality of the  
30  
31 analytical results caused by human errors.  
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36 Protection of the quality of analytical results by managing potential risks and  
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38 mitigation of their severity is an important task for the quality system of any laboratory  
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40 [17]. A laboratory quality system should answer this requirement using components  $j =$   
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42  $1, 2, \dots, J$ , such as validation of the analytical method, training the staff, quality control,  
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44 supervision, etc. Evaluation of possible reduction  $r_{ij}$  of likelihood and severity of human  
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46 error scenario  $i$  as a result of error blocking by quality system component  $j$  is made by  
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48 the same expert(s) using the same scale (0, 1, 3, 9). Notice that blocking human error  
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50 according to scenario  $i$  by quality system component  $j$  can be more effective in the  
51  
52 presence of another component  $j'$  ( $j' \neq j$ ) because of their synergy  $\Delta_{jj'}^{(i)}$ , which is equal to  
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54 0 when the synergy is absent, and equal to 1 when it exists. For example, training ( $j = 2$ )  
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3 is more effective against any error scenario  $i$  when the analytical method is validated  
4 already and a corresponding standard operation procedure is formulated ( $j' = 1$ ). Thus,  
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6  
7  $\Delta_{21}^{(i)} = 1$ . The average synergy factor for quality system component  $j$  is  $s_{ij} = 1 +$   
8  $\sum_{j' \neq j}^J \Delta_{jj'}^{(i)} / (J - 1)$ ,  $1 \leq s_{ij} \leq 2$ . Therefore, reduction of the likelihood and severity of  
9  
10 error scenario  $i$  by quality system component  $j$  (i.e., the risk reduction [15]) is given by  
11  
12  $\tilde{r}_{ij} = r_{ij}s_{ij}$ . In general, there is an  $I \times J$  interrelationship matrix of  $\tilde{r}_{ij}$  values.  
13  
14 Effectiveness score  $Eff^* = (100\%/9) \sum_{j=1}^J \sum_{i=1}^I p_i l_i \tilde{r}_{ij} / \sum_{j=1}^J \sum_{i=1}^I p_i l_i s_{ij}$  of the quality  
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16 system, as a whole, against human errors was formulated in [16] in comparison to an  
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18 ideal quality system with the maximal  $r_{ij} = 9$  for all  $i$  and  $j$ .  
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25 The influence of the variability of expert judgments on the score values was studied  
26 using Monte Carlo simulations of the judgments performed by an R code [18]. The  
27 simulations were based on modeling the expert behavior by means of different  
28 probability mass functions (pmfs) of the expert judgments, i.e., the expert choices on the  
29 scale (0, 1, 3, 9). In particular, reasonably doubting expert judgments were modelled  
30 using a pmf of a chosen value equal to 0.70 and a pmf of close values on the scale in  
31 total equal to 0.30. For example, a judgment equal to 3 from the scale, made by a  
32 reasonably doubting expert, was modelled by a pmf equal to 0.70 at 3, and 0.15 at both  
33 1 and 9 - the closest to 3 values on the scale. It was shown that robustness of the  
34 discussed scores is satisfactory for the quality risk management and improvement of the  
35 laboratory quality system against human errors.  
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49 The technique for quantification of human errors in a chemical analytical laboratory  
50 was successfully applied for pH measurement of groundwater [16], elemental analysis  
51 of geological samples by inductively coupled plasma mass spectrometry (ICP-MS) [18],  
52 and multi-residue analysis of pesticides in fruits and vegetables [19].  
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### 1.3. Aim

The subject of the present paper is evaluating residual risk of human error in chemical analysis (remaining after the error reduction by the laboratory quality system [15]) and quantifying its consequences for quality of analytical results. A procedure for evaluation of the contribution of the residual risk to the measurement uncertainty budget is proposed.

## 2. Residual risk of human errors and its consequences

### 2.1. Quantification

A value of risk reduction  $\tilde{r}_{ij}$  can be normalized by dividing its multipliers  $r_{ij}$  and  $s_{ij}$  by their maximal values, 9 and 2, respectively. Averaging the normalized risk reduction values for the interrelationship matrix (over all the error scenarios and quality system components) leads to score  $r^*$  characterizing the (mean) risk reduction by the laboratory quality system, expressed in %:

$$r^* = (100 \% / 18IJ) \sum_{j=1}^J \sum_{i=1}^I \tilde{r}_{ij} . \quad (1)$$

Then, a score of residual risk of human errors, which are not prevented/blocked or reduced/mitigated by the quality system, is

$$R^* = 100 \% - r^* . \quad (2)$$



The percentage (%) of the quality of analytical results which may be lost due to residual risk of human errors is

$$f_{\text{HE}} = (P^*/100\%) (L^*/100\%) R^* . \quad (3)$$

When a laboratory quality system is able to prevent or block human errors completely, one has  $R^* = 0\%$  and  $f_{\text{HE}} = 0$ : there is no loss of quality, the quality system is ideal and its effectiveness score is  $Eff^* = 100\%$ . If a quality system is not effective at all (or absent),  $Eff^* = 0\%$  since  $\tilde{r}_{ij} = 0$  for all  $i$  and  $j$ . Then  $R^* = 100\%$  and  $f_{\text{HE}} = (P^*/100\%) L^*$ . The extreme case of a complete loss of quality is theoretically possible when, in absence of a quality system ( $R^* = 100\%$ ), scores  $P^*$  and  $L^*$  reach also 100%, i.e., human errors are inevitable and destructive. Thus,  $f_{\text{HE}} = 100\%$  as well.

## 2.2. Distribution of possible loss of quality

Distributions of  $f_{\text{HE}}$  values were studied with Monte Carlo simulations for the model of reasonably doubting expert judgments, analogous to that presented in work [18]. The three sets of expert judgments on human errors in different analytical methods published in papers [16,-18-19] were used here as examples. The values of scores  $P^*$ ,  $L^*$  and  $R^*$ , as well as the results of  $f_{\text{HE}}$  calculations and simulations on a base of 100000 Monte Carlo trials, are presented in Table 1. The  $f_{\text{HE}}$  values calculated directly by formula (3) are close for the three examples. These values can be interpreted as obtained from completely confident expert judgments. A Dirac delta function is applied for modelling pmf of such a judgment: pmf of an expert choice on the scale (0, 1, 3, 9) by this model

Table 1

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3 is 1.00, being 0.00 in total at the rest part of this scale. The mean and median  $f_{HE}$  values  
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5 obviously coincide in such a case, the standard deviation (STD) of the simulated values  
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7 being zero. For all examples, the mean of the simulated  $f_{HE}$  values for the model of  
8  
9 Fig. 1 reasonably doubting expert judgments is a little larger than the  $f_{HE}$  calculated directly  
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11 (not more than for one STD of the simulated values). In other words, the estimated  
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13 possible loss of quality due to residual risk of human errors is larger when an expert  
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15 doubt is taken into account. A similar effect was noted in paper [18] concerning scores  
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17 of likelihood, severity and quality system effectiveness: less confident expert judgments  
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19 lead to less optimistic score values. Histograms of  $f_{HE}$  simulated values are shown in  
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21 Fig. 1a for pH measurement of groundwater, in Fig. 1b – for elemental analysis of  
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23 geological samples with ICP-MS, and in Fig. 1c – for pesticide residue analysis in fruits  
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25 and vegetables. These histograms are practically symmetric, their mean and median  
26  
27 values differing insignificantly. Relative standard deviation values (STD/mean) are in  
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29 the range 0.12 - 0.15, i.e., smaller than 0.4. The same is true if one compares the  
30  
31 maximal difference between the  $f_{HE}$  calculated directly by formula (3) and the mean of  
32  
33 the simulated values with their common average. In Table 1 it is the case of ICP-MS,  
34  
35 where  $f_{HE} = 8.1\%$  by formula (3), while the simulated mean  $f_{HE}$  is  $9.5\%$ , and their  
36  
37 average is  $(8.1\% + 9.5\%)/2 = 8.8\%$ . Since  $(9.5 - 8.1)/8.8 = 0.15 < 0.4$ , one can  
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39 conclude that the  $f_{HE}$  estimates are robust enough to variability of corresponding expert  
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41 judgments [18].  
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### 50 **3. Contribution of human errors to measurement uncertainty budget**

#### 51 *3.1. Evaluation*

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3 Considering the combined uncertainty  $u_c$ , evaluated according to guides [3, 4], as a  
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5 quality parameter of an analytical result, one can say that quality  $Q$  is better, when  $u_c$  is  
6  
7 smaller, i.e.,  $Q = 1/u_c$ . This is the simplest model  $Q(u_c)$  and its simplicity is the main  
8  
9 model advantage. More complicated models could be also investigated and applied in  
10  
11 specific cases.  
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13  
14 Possible loss of quality because of residual risk of human errors is  $Q f_{HE}/100$  % (an  
15  
16 absolute value). Therefore, the resulting quality according to the proposed model is  
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$$20 \quad Q_{res} = Q - Q f_{HE}/100 \% = (1/u_c)(1 - f_{HE}/100 \%). \quad (4)$$

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24 Since  $Q_{res} = 1/u_{cHE}$ , where  $u_{cHE}$  is the combined uncertainty including the human error  
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26 contribution, from formula (4) one has  
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$$30 \quad u_{cHE} = u_c / (1 - f_{HE}/100 \%). \quad (5)$$

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36 In the view of guide [4, pp. 24-25] concerning uncertainty evaluation based on  
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38 judgment “as for standard deviations derived by other methods”, the contribution of the  
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40 uncertainty  $u_{HE}$  caused by residual risk of human errors into the uncertainty budget can  
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42 be approximated by  
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$$45 \quad u_{cHE} = (u_{HE}^2 + u_c^2)^{1/2}. \quad (6)$$

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49 Thus, it follows from formulas (5) and (6) that  
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$$53 \quad u_{HE} = u_c [(1 - f_{HE}/100 \%)^{-2} - 1]^{1/2}. \quad (7)$$

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3 When  $f_{\text{HE}} = 0 \%$ , the uncertainty contribution due to human errors  $u_{\text{HE}} = 0$  and  $u_{\text{CHE}} = u_{\text{c}}$ .

4  
5 When  $f_{\text{HE}}$  increases in the range  $0 \% < f_{\text{HE}} < 100 \%$ , values of  $u_{\text{HE}}$  increase also as shown

6  
7 Fig. 2

8 In Fig. 2. In particular,  $u_{\text{HE}}$  achieves  $1/3u_{\text{c}}$  at  $f_{\text{HE}} = 5 \%$  and begins to be a significant  
9 component of the uncertainty budget by formula (6). At  $f_{\text{HE}} = 68 \%$ , value  $u_{\text{HE}} = 3u_{\text{c}}$   
10 dominates already in the budget: this point is indicated by dotted lines in Fig. 2. When  
11  $f_{\text{HE}}$  exceeds  $68 \%$ ,  $u_{\text{HE}}$  increases with  $f_{\text{HE}}$  dramatically. In the theoretical case of  
12  $f_{\text{HE}} = 100 \%$  formulas (5) and (7) tend to infinity. However, such a contribution of  
13 human error to uncertainty is not realistic, inasmuch as the error becomes apparent: it  
14 will be identified and treated.  
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23 It is known that the largest contribution/component of a combined uncertainty needs  
24 to be investigated more thoroughly [4, p. 49]. Such a contribution may be overestimated  
25 and, hence, simply improved after investigation, or be a subject of a corrective action  
26 requiring an investment. Identified human errors can usually be reduced [20]. Thus, a  
27 good risk management result is when human errors are treated enough by the quality  
28 system to avoid their dominance in the measurement uncertainty budget.  
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### 38 3.2. Examples

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42 The combined standard uncertainty  $u_{\text{c}}$  presented in Table 1 is evaluated, respectively, for  
43 test item preparation for proficiency testing of pH measurement of groundwater [21]  
44 (pH units), for determination of  $10 \text{ ng g}^{-1}$  of  $^{60}\text{Ni}$  in aqueous samples by ICP-MS [22],  
45 and for multi-residue analysis of pesticides in fruits and vegetables (averaged for all  
46 analytes and expressed in % of an analytical result) [23]. Results of calculations of  
47 human error contributions to the measurement uncertainty budget are also presented in  
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56 Table 1.  
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3 The  $f_{\text{HE}}$  values obtained directly by formula (3) were used for calculation of  $u_{\text{HE}}$  and  
4  $u_{\text{CHE}}$  by formulas (6) and (7). At the same time, the mean of the simulated  $f_{\text{HE}}$  values  
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6  
7 allow to examine which uncertainty contribution  $u_{\text{HE}}$  can be obtained if another expert  
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10 will participate in the elicitation process (with different knowledge and experience and,  
11  
12 as a result, with different confidence of judgments). For example, using the mean  $f_{\text{HE}}$   
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14 obtained from judgments of a reasonably doubting expert in ICP-MS leads to  
15  
16  $u_{\text{HE}} = 0.35 \text{ ng g}^{-1}$  and  $u_{\text{CHE}} = 0.83 \text{ ng g}^{-1}$ , which are very close to the  $u_{\text{HE}}$  and  $u_{\text{CHE}}$  values in  
17  
18 Table 1.  
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21 From Table 1, one can understand also that the human error contributions to the  
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23 measurement uncertainty budget in the examples were not negligible. However, it is  
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25 important that these contributions were not dominant and could not influence seriously  
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27 the combined uncertainty. This is a good risk management result.  
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### 32 *3.3. Specificity*

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36 In general, the residual risk of human errors and the corresponding contribution to  
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38 measurement uncertainty may be different in different laboratories active in the same  
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40 field and using the same analytical method. On the other hand, it is impossible to expect  
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42 an equal risk of human errors in chemical analysis by different analytical methods even  
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44 in the same laboratory.  
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47 Changes in the laboratory environment, as well as in any quality system component  
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49 and staff require re-evaluation of the quality of analytical results which may be lost due  
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51 to residual risk of human errors  $f_{\text{HE}}$  and corresponding combined uncertainty  $u_{\text{CHE}}$ . The  
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53 re-evaluation result may indicate either  $f_{\text{HE}}$  increase (e.g., due to retirement of an  
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55 experienced supervisor and/or manager) or its decrease (e.g., due to acquisition of a new  
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3 more accurate and more automated analytical instrument and/or a laboratory  
4 information management system - LIMS) as shown schematically in Fig. 3. The current  
5  $f_{\text{HE}}$  value is demonstrated here by straight line 1 as a result of balance between human  
6 error scenarios  $i = 1, 2, \dots, I$  and quality system components  $j = 1, 2, \dots, J$  blocking the  
7 errors and mitigating their severity. The increased and decreased  $f_{\text{HE}}$  values are  
8 demonstrated by straight lines 2 and 3, respectively. The  $f_{\text{HE}}$  change will not appear  
9 immediately, a certain time  $t$  is necessary for that, as a rule. This process is indicated by  
10 smooth dotted lines connecting the straight lines. Consequently, the combined  
11 uncertainty values  $u_{\text{CHE}}$  will also change according to formulas (6) and (7).  
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#### 25 **4. Conclusion**

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29 In spite of the delicacy of the topic of human errors in chemical analytical laboratories  
30 and a certain misunderstanding of its importance for metrology, this topic should be  
31 discussed without any fear of compromising a laboratory's reputation, just as  
32 measurement uncertainty is discussed. Nowadays, there are no databases of human  
33 errors in different chemical analytical methods. However, experts in these methods have  
34 accumulated the necessary information. Their knowledge and experience may be  
35 quantified using an appropriate scale of expert judgments. Evaluating the residual risk  
36 of human errors in chemical analysis, remaining after the error reduction by a laboratory  
37 quality system, and quantifying the consequences of this risk for quality of analytical  
38 results are possible on the basis of relevant expert judgments. We hope that the  
39 procedure proposed in the present paper for evaluation of the contribution to the  
40 uncertainty budget due to residual risk of human errors will be helpful for a more  
41 complete vision and evaluation of measurement uncertainty in chemical analysis.  
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## References

- [1] JCGM 200:2012. International Vocabulary of Metrology – Basic and General Concepts and Associated Terms, [http://www.bipm.org/utis/common/documents/jcgm/JCGM\\_200\\_2012.pdf](http://www.bipm.org/utis/common/documents/jcgm/JCGM_200_2012.pdf)
- [2] I. Kuselman, F. Pennechi, A. Fajgelj, Y. Karpov. Human errors and reliability of test results in analytical chemistry. *Accred. Qual. Assur.* **18**:3-9 (2013)
- [3] JCGM 100:2008. Evaluation of Measurement Data – Guide to the Expression of Uncertainty in Measurement, [http://www.bipm.org/utis/common/documents/jcgm/JCGM\\_100\\_2008\\_E.pdf](http://www.bipm.org/utis/common/documents/jcgm/JCGM_100_2008_E.pdf).
- [4] S.L.R. Ellison and A. Williams (eds), Eurachem/CITAC Guide: Quantifying uncertainty in analytical measurement, 3<sup>rd</sup> edn., 2012.
- [5] S.L.R. Ellison. Implementing measurement uncertainty for analytical chemistry: the Eurachem Guide for measurement uncertainty. *Metrologia* **51**/4:S199-S205 (2014).
- [6] J.E.T. Andersen. On the development of quality assurance. *Trends in Anal. Chem.* **60**:16-24 (2014).

- 1  
2  
3 [7] I. Kuselman, A. Fajgelj. Human errors and out-of-specification test results. *Chem.*  
4  
5 *Int.* **35**/3:30-31 (2013).  
6  
7 [8] Automotive Industry Action Group (AIAG). Measurement Systems Analysis  
8  
9 Reference Manual, 3rd edn. (2002).]  
10  
11 [9] B. Berglund, G.B. Rossi, J. T. Townsend and L. R. Pendrill (eds), Measurement  
12  
13 with Persons: Theory, Methods, and Implementation Areas. Psychology Press,  
14  
15 New York, 2011.  
16  
17 [10] ISO/IEC 17025:2005. General requirements for the competence of testing and  
18  
19 calibration laboratories.  
20  
21 [11] International Conference on Harmonization (ICH) of Technical Requirements for  
22  
23 Registration of Pharmaceuticals for Human Use. ICH Harmonized Tripartite  
24  
25 Guidelines Q9:2005. Quality risk management.  
26  
27 [12] ISO/TS 22367:2008. Medical laboratories – Reduction of error through risk  
28  
29 management and continual improvement.  
30  
31 [13] U.S. Food and Drug Administration (FDA). Draft Guidance for Industry and FDA  
32  
33 Staff. Applying Human Factors and Usability Engineering to Optimize Medical  
34  
35 Device Design (2011).  
36  
37 [14] ISO 10012:2003. Measurement management systems — Requirements for  
38  
39 measurement processes and measuring equipment.  
40  
41 [15] ISO Guide 73:2009. Risk management — Vocabulary.  
42  
43 [16] I. Kuselman, E. Kardash, E. Bashkansky, F. Pennechi, S.L.R. Ellison, K.  
44  
45 Ginsbury, M. Epstein, A. Fajgelj, Y. Karpov, House-of-security approach to  
46  
47 measurement in analytical chemistry: quantification of human error using expert  
48  
49 judgments, *Accred. Qual. Assur.* **18**:459-467 (2013).  
50  
51 [17] ISO 31000:2009. Risk management — Principles and guidelines.  
52  
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3 [18] I. Kuselman, F. Pennechi, M. Epstein, A. Fajgelj, S. L. R. Ellison. Monte Carlo  
4  
5 simulation of expert judgments on human errors in chemical analysis – a case  
6  
7 study of ICP-MS. *Talanta* **130**:462-469 (2014).  
8  
9  
10  
11 [19] I. Kuselman, P. Goldshlag, F. Pennechi. Scenarios of human errors and their  
12  
13 quantification in multi-residue analysis of pesticides in fruits and vegetables.  
14  
15 *Accred. Qual. Assur.* **19**:361-369 (2014).  
16  
17  
18 [20] IEC/ISO 31010:2009. Risk management — Risk assessment techniques.  
19  
20  
21 [21] E. Kardash, I. Kuselman, I. Pankratov, S. Elhanany. Proficiency testing of pH and  
22  
23 electrolytic conductivity measurements of groundwater: a case study of a  
24  
25 difference between consensus and metrologically traceable values. *Accred. Qual.*  
26  
27 *Assur.* **18**:373-381 (2013).  
28  
29  
30 [22] V.J. Barwick, S.L.R. Ellison, B. Fairman. Estimation of uncertainties in ICP-MS  
31  
32 analysis: a practical methodology. *Anal. Chim. Acta* **394**:281-291 (1999).  
33  
34  
35 [23] I. Kuselman, P. Goldshlag, F. Pennechi, C. Burns. Multi-component out-of-  
36  
37 specification test results: a case study of concentration of pesticide residues in  
38  
39 tomatoes. *Accred. Qual. Assur.* **16**:361-367 (2011)  
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**Figure captions**

**Fig. 1.** Histograms of simulated values  $f_{HE}/\%$  of possible quality loss due to residual risk of human errors for a) pH measurement of groundwater, b) ICP-MS elemental analysis of geological samples, and c) pesticide residue analysis in fruits and vegetables.

**Fig. 2.** Ratio  $u_{HE}/u_c$  of the uncertainty contribution due to residual risk of human errors to the combined uncertainty in dependence on the quality loss  $f_{HE}/\%$ . The case  $u_{HE} = 3u_c$  is indicated by the dotted lines.

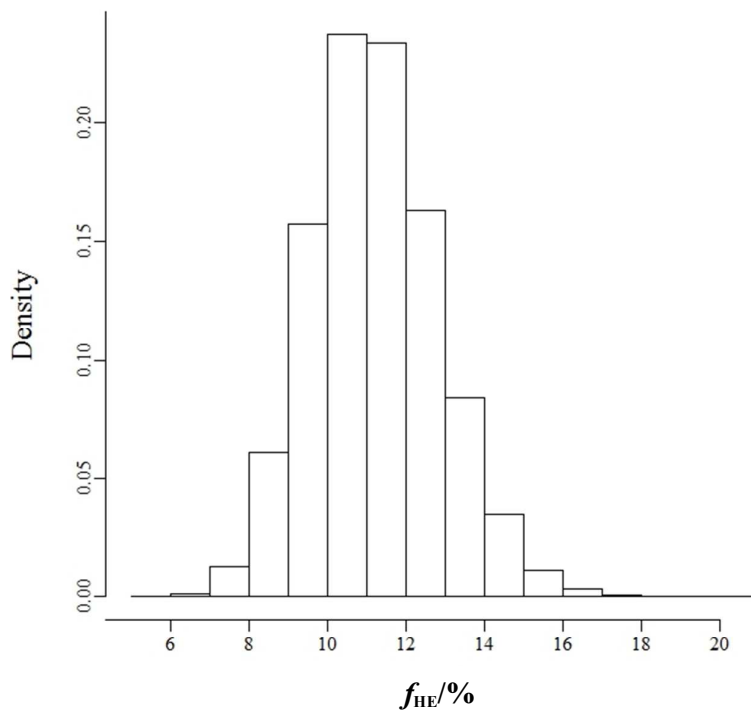
**Fig. 3.** A scheme of possible changes of the quality lost due to residual risk of human errors  $f_{HE}/\%$  vs time  $t/\text{day}$ . The current  $f_{HE}$  value is shown by straight line 1 as a result of the balance between human error scenarios  $i = 1, 2, \dots, I$  and quality system components  $j = 1, 2, \dots, J$  blocking the errors and mitigating their severity. The increased and decreased  $f_{HE}$  values are demonstrated by straight lines 2 and 3, respectively. The process of  $f_{HE}$  changing is indicated by smooth dotted lines connecting the straight lines.

**Table 1.** Evaluation of quality loss due to residual risk of human errors and corresponding contribution to the uncertainty budget

Analytical method	$P^*$ %	$L^*$ %	$R^*$ %	$f_{HE}$ %	MC simulation of $f_{HE}/\%$			$u_c$	$u_{HE}$	$u_{CHE}$
					Mean	Median	STD			
pH metry of groundwater	26	67	62	10.8	11.2	11.1	1.6	0.10	0.05	0.11
ICP-MS of geo-samples	22	56	65	8.1	9.5	9.4	1.4	0.75 ng g <sup>-1</sup>	0.32 ng g <sup>-1</sup>	0.82 ng g <sup>-1</sup>
Pesticide residues in fruits	19	84	63	9.9	10.4	10.4	1.3	20 %	10 %	22 %

Note: STD is the standard deviation of the simulated values from their mean.

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**Fig. 1a**

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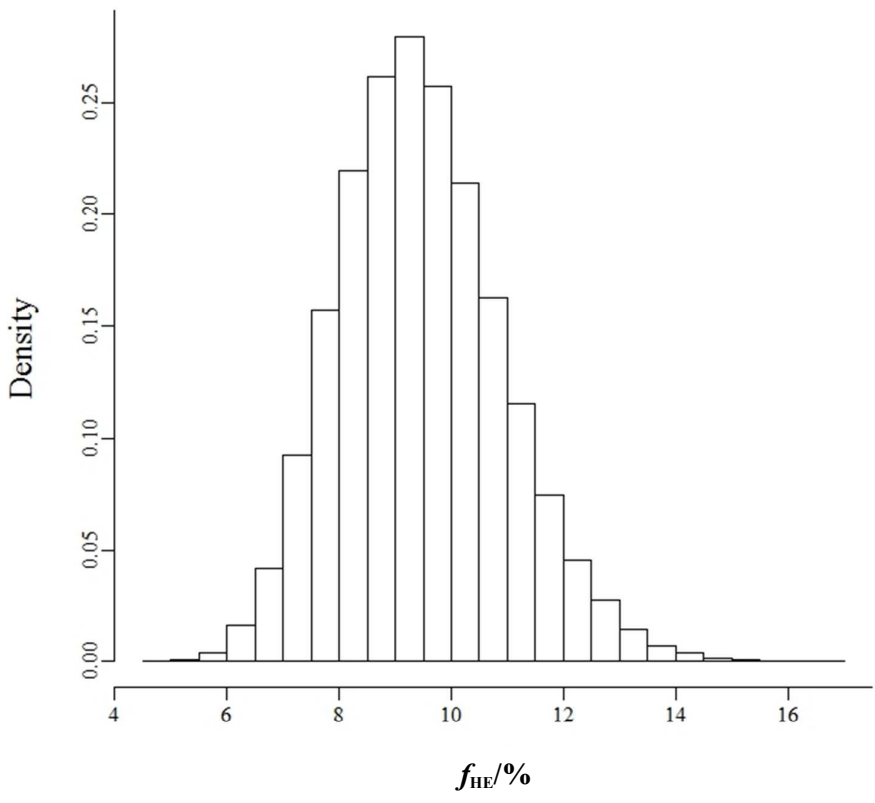


Fig. 1b

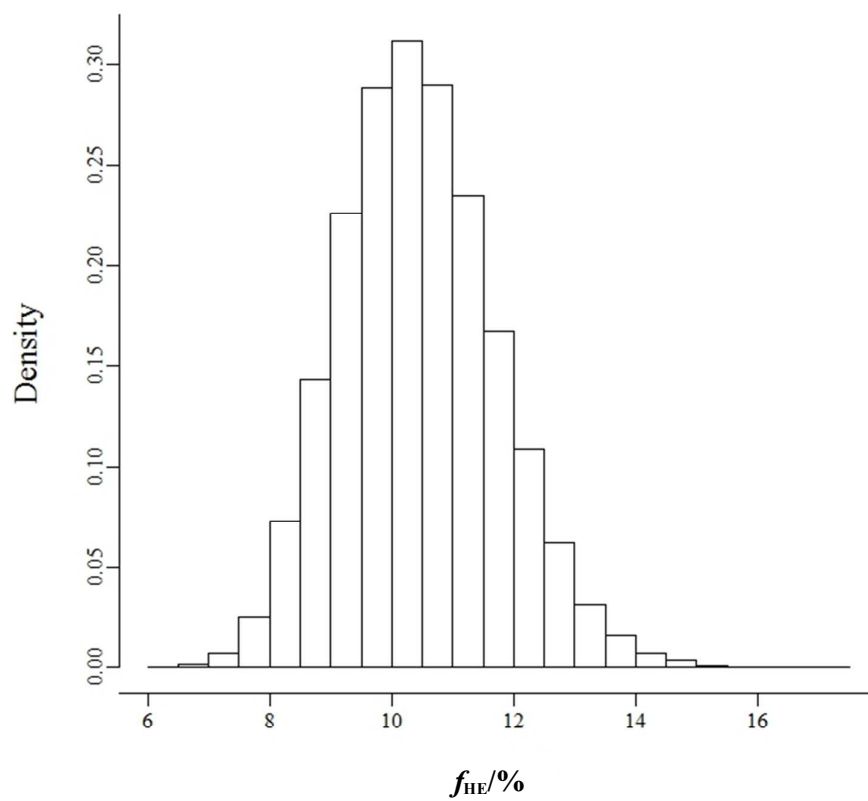


Fig. 1c

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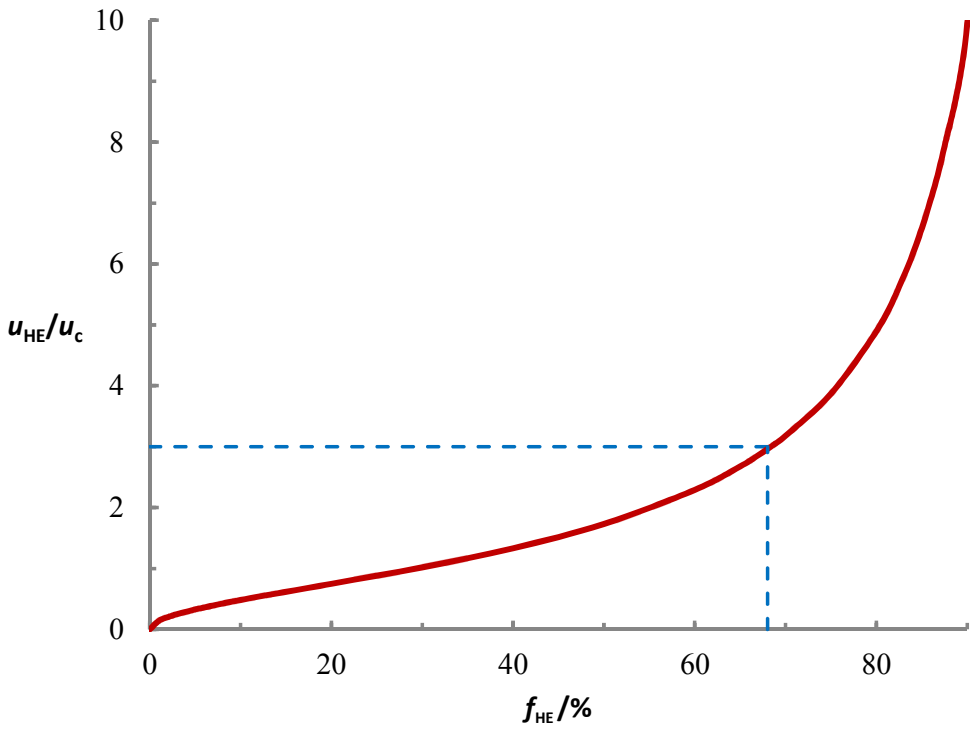


Fig. 2

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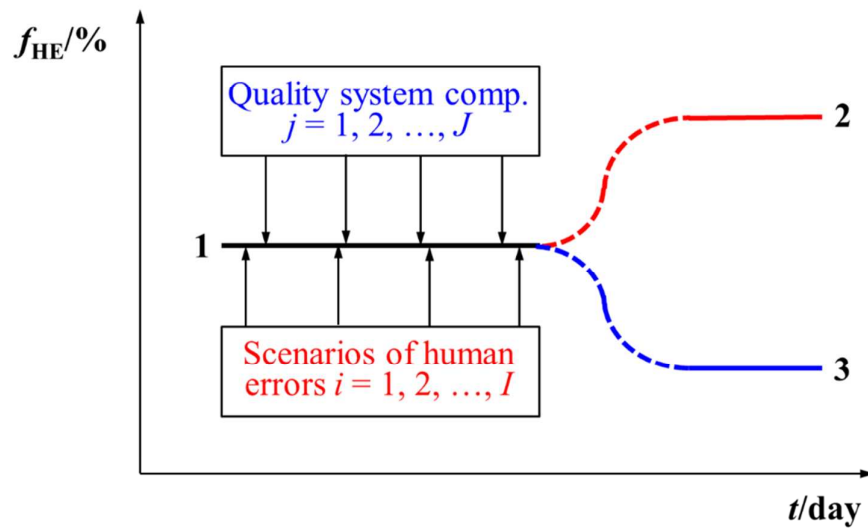


Fig. 3