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This is the author's submitted version of the contribution published as:

Original

Shape-engineered titanium dioxide nanoparticles (TiO₂-NPs): cytotoxicity and genotoxicity in bronchial epithelial cells / Gea, Marta; Bonetta, Sara; Iannarelli, Luca; Giovannozzi, Andrea Mario; Maurino, Valter; Bonetta, Silvia; Hodoroaba, Vasile-Dan; Armato, Caterina; Rossi, Andrea Mario; Schilirò, Tiziana. - In: FOOD AND CHEMICAL TOXICOLOGY. - ISSN 0278-6915. - 127:(2019), pp. 89-100. [10.1016/j.fct.2019.02.043]

Availability:

This version is available at: 11696/61722 since: 2021-03-09T19:22:40Z

Publisher:

Elsevier

Published

DOI:10.1016/j.fct.2019.02.043

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(Article begins on next page)

Food and Chemical Toxicology
Manuscript Draft

Manuscript Number: FCT-D-18-02775

Title: Shape-engineered titanium dioxide nanoparticles (TiO₂-NPs): cytotoxicity and genotoxicity in bronchial epithelial cells

Article Type: Full Length Article

Keywords: shape-engineered TiO₂ nanoparticles; genotoxic and oxidative damage; Comet assay; cytotoxicity; Raman spectroscopy

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In conclusion, cytotoxicity of NPs was low, influenced by shape as well as by light exposure. Instead, genotoxicity seemed to be influenced by cellular-uptake and aggregation tendency. This study suggest that shape engineered TiO₂-NPs are safer than the commercial ones.

Shape-engineered titanium dioxide nanoparticles (TiO₂-NPs): cytotoxicity and genotoxicity in bronchial epithelial cells

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^a Department of Public Health and Pediatrics, University of Turin, Piazza Polonia 94, 10126 Turin, Italy;

^b Quality of Life Division, Istituto Nazionale di Ricerca Metrologica, Strada delle Cacce 91, 10135 Turin, Italy;

^c Department of Chemistry, University of Turin, Via Giuria 7, 10125 Turin, Italy;

^d Surface Analysis and Interfacial Chemistry division, Federal Institute for Materials Research & Testing (BAM), 12200 Berlin, Germany;

^e Centre for Sustainable Future Technologies (CSFT@PoliTo), Istituto Italiano di Tecnologia, Corso Trento 21, 10129 Turin, Italy;

***Corresponding author:**

Sara Bonetta

Department of Public Health and Pediatrics,

University of Torino,

Piazza Polonia 94, 10126 Turin, Italy,

Tel: +390116708192

e-mail address: sara.bonetta@unito.it

Highlights

Shape-engineered titanium dioxide nanoparticles (TiO₂-NPs): cytotoxicity and genotoxicity in bronchial epithelial cells

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Highlights

- Cytotoxicity/genotoxicity evaluation of engineered TiO₂-NPs with different shapes
- TiO₂-NPs cytotoxicity was low, influenced by the shape and by light exposure
- Genotoxicity was influenced by cellular-uptake and aggregation tendency of TiO₂-NPs
- The presence of light enhanced the oxidative DNA damage
- It seems that shape engineered TiO₂-NPs are safer than the commercial ones

Shape-engineered titanium dioxide nanoparticles (TiO₂-NPs): cytotoxicity and genotoxicity in bronchial epithelial cells

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^c[Department of Chemistry, University of Turin, Via Giuria 7, 10125 Turin, Italy](#);

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^eCentre for Sustainable Future Technologies (CSFT@PoliTo), Italian Institute of Technology, Corso Trento 21, 10129 Turin, Italy;

***Corresponding author:**

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Department of Public Health and Pediatrics,

University of Torino,

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List of Abbreviations

BEAS-2B – Human bronchial epithelial cells

D_h – Hydrodynamic diameter

Fpg – formamidopyrimidine glycosylase

NP – Nanoparticle

TiO_2 – Titanium dioxide

Abstract

The aim of this study was to evaluate cytotoxicity (WST-1 and LDH assays) and genotoxicity (Comet assay) of three engineered TiO₂-NPs with different shapes (bipyramids, rods, platelets) in comparison with two commercial TiO₂-NPs (P25, food grade).

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2 **genotoxicity in bronchial epithelial cells**

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12 6 Schilirò^a

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16
17 8 ^a Department of Public Health and Pediatrics, University of Turin, Piazza Polonia 94, 10126
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19 9 Turin, Italy;

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22 10 ^bQuality of Life Division, National Institute of Metrological Research, Strada delle Cacce 91,
23
24 11 10135 Turin, Italy;

25
26
27 12 ^cDepartment of Chemistry, University of Turin, Via Giuria 7, 10125 Turin, Italy;

28
29 13 ^dSurface Analysis and Interfacial Chemistry division, Federal Institute for Materials Research
30
31
32 14 & Testing (BAM), 12200 Berlin, Germany;

33
34 15 ^eCentre for Sustainable Future Technologies (CSFT@PoliTo), Italian Institute of
35
36
37 16 Technology, Corso Trento 21, 10129 Turin, Italy;

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40
41 18 ***Corresponding author:**

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43
44 19 Sara Bonetta

45
46 20 Department of Public Health and Pediatrics,

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48 21 University of Torino,

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50
51 22 Piazza Polonia 94, 10126 Turin, Italy,

52
53 23 Tel: +390116708192

54
55
56 24 e-mail address: sara.bonetta@unito.it

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28	D _h – Hydrodynamic diameter
29	Fpg – formamidopyrimidine glycosylase
30	NP – Nanoparticle
31	TiO ₂ – Titanium dioxide
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76 **1. Introduction**

77 Nanoparticles (NPs) are defined as particles having their three dimension in the range of 1 –
78 100 nm (ISO/TS 27687:2008). Actually, many consumer products incorporates NPs. The
79 technological, medical and economic benefits of NPs are considerable, but the presence of
80 nanoparticles in the environment could cause adverse effects to humans. NPs have a grater
81 surface area per unit mass, so they potentially have an increased biological activity compared
82 to fine particles. Moreover, NPs size is comparable to size of cellular structures, so NPs
83 might potentially emulate biological molecules or interfere physically with biological
84 processes (Magdolenova *et al.* 2012a).

85 TiO₂ is the oxide of titanium and it has different crystalline structures: anatase, brookite and
86 rutile. Brookite is not produced by industry and is not incorporated in commercial products.

87 In contrast, rutile and anatase are largely used in commercial products (Jovanovic 2015).

88 TiO₂ is one of the most frequently applied NPs and it is in the top five NPs used in consumer
89 products (Shi *et al.* 2013). TiO₂-NPs produced are used primarily as a pigment owing to their
90 brightness, resistance to discoloration and high refractive index. As a pigment TiO₂-NPs are
91 incorporated in paints, plastic materials, paper, foods, medical products and cosmetics. Due
92 to its catalytic and photocatalytic properties, TiO₂ is also used as an antimicrobial agent and a
93 catalyst for purification of air and water (Bonetta *et al.* 2013, Tomankova *et al.* 2015).

94 TiO₂-NPs could be engineered in terms of shapes and sizes by changing synthesis conditions
95 such as raw material, temperature, acidic and alkaline conditions. Engineered TiO₂-NPs with
96 various shapes (e.g. rods, dots and belts) have been prepared for different applications
97 (Bernard and Curtiss 2005, Sha *et al.* 2015, Wang *et al.* 2004). In particular engineered fiber-
98 shaped nanomaterials (i.e. nanowires, nanotubes) are very attractive because showed higher
99 activity and advantages in photocatalysis, charge transfer and sensing applications due to

100 their structure (Hamilton *et al.* 2009). However, these new and enhanced properties may also
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3 101 induce higher toxicological effects upon exposure with biological tissues.

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5 102 Humans can be exposed to TiO₂-NPs via three portals of entry: oral (mainly via food
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7 103 consumption), dermal (often through cosmetic and sunscreen applications) and inhalation
8
9 104 (mainly under occupational and manufacturing conditions) (Warheit and Donner 2015).

10
11 105 Based on the evidence that TiO₂ can induce lung cancer in rats, TiO₂-NPs were classified as
12
13 106 possibly carcinogenic to humans (group 2B) by the International Agency for Research on
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15 107 Cancer (IARC; Kuempel and Ruder 2012). Indeed, the inhalation and instillation of rutile and
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17 108 anatase TiO₂-NPs induced lung tumors (Xu *et al.* 2010), broncho-alveolar adenomas and
18
19 109 cystic keratinizing squamous cell carcinomas (De Matteis *et al.* 2016; Mohra *et al.* 2006).
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21 110 TiO₂-NPs were also classified as potential occupational carcinogens by the National Institute
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23 111 for Occupational Safety and Health (NIOSH 2011; Chen *et al.* 2014).

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27 112 Many *in vitro* studies showed cytotoxicity, genotoxicity and oxidative effects induced by
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29 113 TiO₂-NPs through oxidants generation, inflammation and apoptosis (Jugan *et al.* 2011,
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31 114 Karlsson *et al.* 2015, Park *et al.* 2008, Shi *et al.* 2010). The potential of NPs to cause DNA
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33 115 damage is an important aspect that needs attention because it could induce mutations and
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35 116 carcinogenesis. Physico-chemical characteristics of NPs have an important role on toxicity.
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37 117 Different studies showed that biological effects can be influenced by crystalline structure,
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39 118 size, shape, exterior area, agglomeration/aggregation and surface properties (Bhattacharya *et*
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41 119 *al.* 2009, Johnston *et al.* 2009). Some studies revealed that crystalline structure probably
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43 120 influences the induced toxicity, in particular the anatase seems to be more reactive (Sayes *et*
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45 121 *al.* 2006) and induces more toxic, genotoxic and inflammatory effects, than the rutile (Falck
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47 122 *et al.* 2009, Petkovic *et al.* 2011, Xue *et al.* 2010). However, other studies gave contradictory
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49 123 results with rutile forms being more toxic than anatase (Gurr *et al.* 2005, Numano *et al.* 2014,
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51 124 Uboldi *et al.* 2016). The effect of agglomeration/aggregation of NPs on toxicity is not well
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125 understood yet. In recent studies, some authors demonstrated that agglomeration can
126 influence NPs genotoxicity (Magdolenova *et al.* 2012b, Prasad *et al.* 2013).

127 Although physico-chemical properties of NPs can have an important role in the impact on
128 their toxicity, only few studies on shape dependent TiO₂ toxicity has been conducted (Allegri
129 *et al.* 2016, Hamilton *et al.* 2009, Park *et al.* 2013). Additional studies are needed to evaluate
130 the role of shape on TiO₂-NPs toxicity in order to produce useful data for assessing the safety
131 of engineered NPs. To address this issue, the aim of this study was to investigate cytotoxicity
132 (WST-1 and LDH) and genotoxicity (Comet assay) of three types of engineered TiO₂-NPs of
133 different shapes (bipyramids, rods and platelet NPs) in BEAS-2B (cells isolated from human
134 bronchial epithelium) in comparison with two commercial types of TiO₂-NPs (P25 and food
135 grade). Since the exposure to TiO₂-NPs mainly occurs through respiratory tract, human cells
136 of the respiratory system (such as BEAS-2B), were selected as a good cell model for *in vitro*
137 toxicology tests. All the TiO₂-NPs in this study were first physico-chemically characterized,
138 even in different culture media to study their agglomeration state, and then they were
139 biologically evaluated. In order to take into account the photocatalytic properties of the TiO₂-
140 NPs, we investigated the effects on cytotoxicity and genotoxicity on BEAS-2B under light
141 exposure and in darkness. Moreover, a modern application of Raman spectroscopy, the 3D
142 confocal Raman imaging, was used to study the uptake of the NPs within the BEAS-2B cells,
143 as the Raman spectra provide information about both organic molecules and solid NPs
144 simultaneously (Ahlinder *et al.* 2013).

146 **2. Materials and methods**

147 **2.1 Synthesis and Preparation of TiO₂ NPs dispersion**

148 Rods and bipyramids TiO₂-NPs were synthesized by the forced hydrolysis of an aqueous
149 solution of TiIV(triethanolamine)₂titanatrane (Ti(TEOAH)₂), using triethanolamine (TEOA)

150 as shape controller; pH of synthesis was adjusted by adding 1 M NaOH solution; details of
151 these procedures were previously reported (Iannarelli *et al.* 2016, Lavric *et al.* 2017). The
152 synthesis of platelet NPs was performed with a solvothermal method (Han *et al.* 2009, Zhang
153 *et al.* 2012). In a typical synthesis: a precise volume of Ti(OBu)₄ was added in a 150 ml
154 Teflon pot and the desired volume of concentrated hydrofluoric acid was added dropwise
155 under stirring. The Teflon pot was sealed and kept under stirring at high temperature (250°C)
156 for 24h in autoclave. The resulting paste was centrifuged three times and washed with
157 acetone and with water (Milli-Q) to remove the residual organics. The synthesis dispersions
158 were subjected to dialysis process (against ultrapure water, using Spectra/Por dialysis
159 membrane tubing MWCO 8–14 kDa) in order to clean the medium. To avoid agglomeration
160 and precipitation, dimethylsulfoxide (DMSO 1% in water) was added to the NPs dispersions
161 and the dispersions were homogenized using an ultra-sonication procedure, as previously
162 described by Iannarelli *et al.* (2016).
163 The same ultra-sonication procedure was employed in the preparation of the dispersion of
164 commercial TiO₂ powders, which were the P25 NPs (Evonik), extensively used in toxicity
165 studies (Karlsson *et al.* 2015, Magdolenova *et al.* 2014, Valant *et al.* 2012), and the food
166 grade NPs (Faravelli Group), incorporated in many edible products (Weir *et al.* 2012).
167 Powders of commercial TiO₂-NPs were first dispersed in a solution of DMSO (1% in water)
168 and then ultrasonicated as described above.
169 Stock solutions of both commercial and engineered TiO₂-NPs were prepared at the final
170 concentration of 2.5 mg/ml.

171 **2.2 Scanning Electron Microscopy (SEM) including Transmission Mode (T-SEM)**

172 The dimensional characterization (size and shape) of TiO₂-NPs was carried out with SEM
173 using a Zeiss Supra 40 instrument (Zeiss) equipped with a Schottky field emitter, the standard
174 secondary electrons, i.e. Everhart-Thornley, detector and a high-resolution In-lens detector.

175 The surface-sensitive In-lens SEM mode better suited to morphological/shape analysis and
176 transmission mode in SEM (T-SEM) better suited for dimensional measurements were
177 applied complementary to the same field of view on the sample.

178 ***2.3 Dynamic Light Scattering (DLS) analysis***

179 Delsa Nano™ C Analyzer (Beckman Coulter) equipped with a 638 nm diode laser and a
180 temperature control was used for the DLS measurements. The laser fluctuation was detected
181 on a photomultiplier tube detector positioned behind the cuvette with an angle of 163°.
182 Hydrodynamic diameters were calculated setting temperature at 25°C, viscosity (η) 0.890 cP
183 and refractive index of water 1.3325. In order to simulate the culture medium conditions,
184 DLS analyses were conducted on dilution of TiO₂ dispersions (1:4) in a 1% DMSO aqueous
185 solution, as reference analysis, and in base and complete RPMI 1640 medium.

186 ***2.4 Raman spectroscopy analysis***

187 The aqueous suspensions of the TiO₂-NPs under investigation were freeze-dried to obtain a
188 solid powder. Raman spectroscopy was used in the analysis of dry TiO₂-NPs powder using a
189 DXR™ Raman Microscope (Thermo Scientific) with a laser wavelength at 532 nm, a laser
190 power of 1 mW and a 10x microscope objective. Spectra were collected in the 50–1800 cm⁻¹
191 spectral region, with a grating resolution of 3.3–3.9 cm⁻¹, exposure time of 1 s and 20 scans in
192 total.

193 ***2.5 Cell culture and exposure***

194 BEAS-2B cells, isolated from human bronchial epithelium, were obtained from the ATCC
195 (American Type Culture Collection). BEAS-2B were grown, maintained and treated in
196 completed RPMI 1640 medium (37°C, 5% CO₂).

197 Just before the exposure the fresh stock solution of NPs (2.5 mg/ml) was vortexed and
198 sonicated (30 min) in order to disperse the NPs. NPs (5 – 160 µg/ml) were added to the cells
199 and mixed on a shaker (10 min). The cells were exposed for 1h under laboratory light

200 (36W/840 Lumilux Cool White-36 W, 3350 lm, 4000 K-supplied from OSRAM lighting AG)

201 and then incubated at 37 °C in darkness (23h) (exposure with light). To quantify effects due

202 to the photocatalytic activity of TiO₂, cells were exposed for 24h in darkness (exposure in

203 darkness).

204 After exposure, cytotoxicity and genotoxicity assays were performed.

205 **2.6 Cytotoxicity**

206 Cell viability was assessed using Cell Proliferation Reagent WST-1 (Roche). The assay was

207 performed as previously described by Gea et al. (2018). Briefly, BEAS-2B were seeded in

208 24-well plates (5×10^4 cells/well) and exposed to NPs (5, 10, 20, 50 and 80 µg/ml, equivalent

209 to 1.3, 2.6, 5.2, 13.0, 20.7 µg/cm²). After exposure, WST-1 was added (50 µl/well) and

210 incubated for 3h (37 °C). After incubation, well contents were centrifuged and the

211 supernatants were transferred in 96-well plate to remove the interference owing to the NPs.

212 The absorbance was measured at 440 nm (Tecan Infinite Reader M200 Pro). Absorbance of

213 unexposed cells was used as negative control. Data were expressed as a percentage of

214 viability. All experiments were performed in quadruplicate (four wells for each experimental

215 condition).

216 As indicator of cell membrane damage, Lactate dehydrogenase activity was measured in cell-

217 free culture supernatants using the LDH assay kit (Cytotoxicity Detection Kit PLUS, Roche)

218 modified for NPs exposure. Briefly, BEAS-2B cells were seeded in 24-well plates (5×10^4

219 cells/well) and exposed to NPs (5, 10, 20, 50 and 80 µg/ml, equivalent to 1.3, 2.6, 5.2, 13.0,

220 20.7 µg/cm²). After exposure, the contents of each well were centrifuged to remove the

221 interference owing to the NPs. Each supernatant (100 µl) was transferred into 96-well plate,

222 mixed with Reaction Mixture (100 µl/well) and incubated for 30 min at 15 – 25 °C. After

223 incubation, Stop Solution (50 µl/well) was added and the absorbance was measured at 490

224 nm (Tecan Infinite Reader M200 Pro). Absorbance measurement of unexposed cells was

225 used as negative control, while absorbance measurement of unexposed lysed cells was used
226 as positive control. LDH release was expressed as a percentage of control cells. All
227 experiments were performed in triplicate (three wells for each experimental condition).

228 **2.7 Genotoxicity**

229 For DNA damage evaluation the BEAS-2B cells were cultured overnight in 6-well plates ($3 \times$
230 10^5 cells/well) before exposure to NPs. Cells were exposed to different doses of NPs: 20, 50,
231 80, 120 and 160 $\mu\text{g/ml}$, equivalent to 5.2, 13.0, 20.8, 31.2, 41.6 $\mu\text{g/cm}^2$. Unexposed cells and
232 cells treated with NPs dispersion liquid (DMSO 1%) were used as negative controls. After
233 exposure, cell viability was determined using trypan blue staining. The Comet assay was
234 performed according to Tice *et al.* (2000) after slight modifications (Bonetta *et al.* 2009). The
235 percentage of tail intensity was used to estimate DNA damage. A hundred randomly selected
236 cells per treatment (2 spot) were analyzed using the Comet Assay IV software (Perceptive
237 Instruments Ltd). Two independent experiments were performed for each experimental
238 condition.

239 The oxidative DNA damage was evaluated using the formamidopyrimidine glycosylase
240 (Fpg)-modified Comet assay as reported in Bonetta *et al.* (2018).

241 For each experimental point, the mean % tail intensity from enzyme untreated cells (direct
242 DNA damage) and mean % tail intensity for Fpg-enzyme treated cells (direct and indirect
243 DNA damage) were calculated. Two independent experiments were performed for each
244 experimental condition.

245 **2.8 3D confocal micro-Raman imaging spectroscopy**

246 Raman grade Calcium fluoride (CaF_2) windows (Crystran Technology srl) were employed as
247 alternative substrate instead of standard plastic substrates for cells growing due to the low
248 toxicity and almost absent background signals (Kann *et al.* 2015). The BEAS-2B cells were
249 cultured overnight in 6-well plates on a CaF_2 substrate (3×10^5 cells/well) before exposure to

250 NPs. Cells were treated with NPs (80 $\mu\text{g/ml}$, 24h). After exposure, cells were washed twice
251 with PBS and fixed with 3 ml of methanol. CaF_2 substrates were dried and stained with
252 Giemsa dye (4% Giemsa's azur eosin methylene blue solution, 4% Sorensen buffer 0.067 M
253 pH 6.8, 8 min at room temperature), then washed twice with distilled water and dried.
254 Giemsa staining is one of the standard procedures in histology, useful to evidence
255 morphological cells features, such as cell nuclei, which appear in various shades of
256 red/purple, and the cytoplasm, which appears blue.

257 3D confocal micro-Raman imaging spectroscopy of BEAS-2B cells was conducted with a
258 DXRTMxi Raman Imaging Microscope (Thermo Scientific) using a laser wavelength at 532
259 nm, a 1 mW laser power, a 100X microscope objective and a motorized stage with a 1 μm of
260 step size and a 1 μm offset. Spectra were collected in the 50–3500 cm^{-1} spectral region with a
261 grating resolution of 5 cm^{-1} , an exposure time of 0.025 s and 5 scans in total. 3D Raman
262 images were reconstructed taking the Raman peaks at 1600 cm^{-1} of methylene blue and the E_g
263 band at 144 cm^{-1} of the TiO_2 -NPs, respectively. Each cell was investigated at different focal
264 planes and a chemical image was obtained by the combination of the $\nu(\text{C-C})$ ring at 1600 cm^{-1}
265 of the methylene blue and the E_g band at 144 cm^{-1} of the TiO_2 -NPs. Since methylene blue
266 is contained in the Giemsa stain and it is widely distributed into the fixed cells, its signals
267 were considered representative of the entire volume of the cells. As far as the tracking of the
268 NPs are concerned, the E_g band at 143 cm^{-1} is the most intense signal in the molecular
269 fingerprint of the anatase TiO_2 and the region between 50 cm^{-1} and 400 cm^{-1} in the Raman
270 spectrum is usually free of the vibrational bands of biological species. Therefore, this signal
271 was selected to sensitively locate the TiO_2 -NPs inside the cells. Image J software was used in
272 the development of the 3D chemical images both for cells and TiO_2 -NPs, which were
273 superimposed using a Solidworks[®] 2016 Cad based software. 3D Raman chemical images
274 are presented using a color meshwork i.e. blue for cell tissues and red for TiO_2 agglomerates.

275

276 **2.9 Statistical analysis**

277 IBM SPSS software (ver. 24.0) was used to perform statistical analysis. The results of WST-
278 1, LDH and Comet assays are presented as the mean±standard deviation. Differences
279 between exposed and control cells were tested by ANOVA followed by Dunnett's test
280 procedure. Differences between light and dark exposure were tested by ANOVA, followed by
281 the Tukey's test procedure. Data were considered statistically different for a p-value less than
282 0.05.

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284 **3. Results**

285 **3.1 Raman characterization of NPs and size distribution**

286 In order to establish a relationship among the physico-chemical features of NPs and their
287 ability to induce a toxic effect, well-defined and controlled protocols were developed for the
288 production of engineered anatase TiO₂-NPs with different shapes. All the NPs produced in
289 this study were first characterized with a SEM equipped with a transmission-unit for T-SEM,
290 which provided information both on the shape and the size of the constituent NPs (Fig. 1a-e).
291 The Fig. 1 and Table 1 show shapes and particle size of commercial TiO₂-NPs and fabricated
292 engineered TiO₂-NPs.

293 These NPs were also characterized by Dynamic Light Scattering (DLS) as a quick method for
294 sizing and determining the state of NP agglomeration. For each kind of sample, the
295 agglomeration in 1% DMSO aqueous solution, in base RPMI and complete RPMI (Fig. 1f-j)
296 were compared. In all the TiO₂ materials considered for this study, the agglomeration state
297 increase in base RPMI, while the size distribution in DMSO and in complete RPMI is quite
298 similar.

299 The crystalline composition of the TiO₂-NPs, analyzed by Raman spectroscopy, showed a
300 typical fingerprint of the anatase TiO₂ (Fig. S.1) with the characteristic phonon bands E_g at
301 143 cm⁻¹, E_g at 197 cm⁻¹, A_{1g} at 397 cm⁻¹, B_{1g} at 515 cm⁻¹ and E_g at 639 cm⁻¹ for all the
302 investigated NPs. Since P25 is a known mixture of anatase and rutile (5:1), with also a small
303 amount of amorphous TiO₂ (Ohtani *et al.* 2010), its Raman spectrum still retains all the
304 typical anatase Raman bands but it also contains two small shoulders at 450 cm⁻¹ and 600 cm⁻¹,
305 which were assigned to the E_g and A_{1g} phonon bands, respectively, of rutile (Tompsett *et al.* 1995). All the physiochemical properties of the TiO₂-NPs under study such as shape,
306 particle size, hydrodynamic diameter in different liquid media and the crystalline phase are
307 summarized in Table 1.

309 **3.2 Cytotoxicity**

310 The results of the effects of different TiO₂-NPs concentration on cell viability (WST-1 assay)
311 are reported in Fig. 2a (exposure with light) and in Fig. 2b (exposure in darkness).

312 In general, a low cytotoxic effect was observed at the tested doses both in the exposure with
313 light and in the exposure in darkness. The observed viability ranged from 102.8 to 88.4% for
314 the exposure with light and from 99.6 to 87.4% for the exposure in darkness.

315 Considering the exposure with light, the commercial P25 induced a slight decrease in
316 viability starting from the doses of 50 µg/ml (p<0.05) while no cytotoxic effects were
317 observed for the other commercial NPs (food grade) at the tested concentrations. As far as
318 engineered NPs are concerned, bipyramids and platelet NPs induced the same cytotoxic
319 effect of commercial P25 NPs; on the contrary, rods is the NP shape with higher cytotoxic
320 effect showing a viability decrease already starting from 10 µg/ml (p<0.05 or p<0.001).

321 Considering the exposure in darkness, a lower cytotoxic effect was observed for commercial
322 P25 NPs with respect to light exposure because a slight decrease in viability was observed for
323 P25 NPs only at the highest dose (80 µg/ml) (p<0.05). As reported after exposure with light,

324 no cytotoxic effect was observed for the other commercial NPs (food grade). About
325 engineered NPs, the exposure in darkness did not modify the cytotoxic effect of bipyrramids
326 NPs resulting in a viability reduction starting from the dose of 50 $\mu\text{g/ml}$ ($p<0.001$) as
327 reported in the experiment with light. In contrast, in the darkness, rods NPs showed a lower
328 cytotoxic effect than observed with light because a slight decrease in viability was observed
329 for rod NPs only starting from the dose of 20 $\mu\text{g/ml}$ ($p<0.05$ or $p<0.001$). As during the
330 exposure with light, platelet NPs induced a decrease in viability; the cytotoxic effect was
331 significant starting from a less dose (10 $\mu\text{g/ml}$, $p<0.05$) than in the experiment with light (50
332 $\mu\text{g/ml}$).

333 The results of the effects of different TiO_2 -NPs concentration on lactate dehydrogenase
334 activity (LDH assay) has been reported in Fig. 2c (exposure with light) and in Fig. 2d
335 (exposure in darkness).

336 No significant signs of cytotoxicity were seen by the LDH assay in both exposure protocols
337 (with light or in darkness), confirming the low cytotoxic effect evidenced by WST-1 assay.

338 **3.3 Genotoxicity**

339 The results of genotoxic effect and oxidative DNA damage induced by different
340 concentration of NPs are reported in Fig. 3.

341 Considering the exposure with laboratory light, no genotoxic effect was showed in enzyme
342 untreated cells (direct DNA damage) for commercial P25 NPs (Fig. 3a). On the other hand, a
343 dose-dependent increase of DNA damage was observed for these NPs in enzyme treated cells
344 (direct and indirect DNA damage) respect to the control cells ($p<0.05$ or $p<0.001$), with the
345 exception of the last dose (160 $\mu\text{g/ml}$) that induced a DNA damage equal to 80 $\mu\text{g/ml}$. A
346 significant oxidative damage was observed for P25 NPs starting from 50 $\mu\text{g/ml}$ ($p<0.05$ or
347 $p<0.001$). The results obtained with the other commercial NPs (food grade)(Fig. 3b) showed
348 the presence of a significant dose-response DNA damage both in enzyme untreated cells and

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349 in enzyme treated cells starting from 50 µg/ml. Moreover, the difference between the two
350 effects resulted significant starting from 20 µg/ml ($p<0.05$ or $p<0.001$) highlighting an
351 oxidative damage induced by food grade NPs.

352 Respect to commercial NPs, engineered NPs showed a lower extent of DNA damage. In
353 particular, neither genotoxic effect nor oxidative damage were observed for engineered
354 bipyramids and rods NPs (Fig. 3c,d). Platelet NPs induced a significant DNA damage respect
355 to the control cells ($p<0.05$ or $p<0.001$) both in enzyme untreated cells and in enzyme treated
356 cells and they induced a significant oxidative DNA damage starting from 80 µg/ml ($p<0.001$)
357 (Fig. 3e). However in contrast with commercial NPs (food grade), a dose-response of the
358 effects were not observed.

359 As demonstrated by other authors (Kalsson 2010, Karlsson *et al.* 2015), an interference
360 during the scoring of the assay was detected in particular at the higher doses of P25 and
361 platelet NPs, indeed nanoparticles with some autofluorescence were visible in the comets
362 “head” and the stained DNA appeared faded. The interference probably caused the loss of
363 concentration-dependent increase in DNA direct and oxidative damage observed for the
364 higher doses. The phenomenon could be explained also considering that base oxidation is
365 hard to measure accurately when there are a lot of strand breaks, because the Comet assay
366 becomes saturated (Collins *et al.* 2017).

367 In order to evaluate the role of the light on the genotoxic and oxidative damage induced by
368 commercial and engineered NPs, the highest doses (80, 120, 160 µg/ml) of NPs that showed
369 a genotoxic effect (P25, food grade and platelet NPs) were tested in darkness (24h).

370 Considering the exposure in darkness, no genotoxic effect was observed for commercial P25
371 NPs in enzyme untreated cells (direct DNA damage) (Fig. 3f) as reported in the experiment
372 with light (Fig. 3a). However, in the enzyme treated cells a dose-response DNA damage
373 (direct and indirect DNA damage) was observed with respect to control cells ($p<0.05$ or

374 p<0.001), but oxidative DNA damage was lower than in the experiment with light (p<0.05 or
375 p<0.001). The commercial food grade NPs induced a significant dose-response DNA damage
376 both in enzyme untreated cells and in enzyme treated cells (p<0.001 and p<0.05 respectively)
377 (Fig. 3g). However, the DNA damage resulted in both cases lower than in the experiment
378 with light p<0.05 or p<0.001) and an oxidative damage was induced only at the highest dose
379 (160 µg/ml) (p<0.05).

380 With regard to engineered NPs, platelet NPs induced a significant DNA damage with respect
381 to the control cells (p<0.05 or p<0.001) both in enzyme untreated cells and in enzyme treated
382 cells (Fig. 3h). However, while the DNA damage in enzyme untreated cells was equivalent to
383 the DNA damage induced in the experiment with light (Fig. 3e), a decrease of DNA damage
384 in enzyme treated cells was observed, resulting in no oxidative damage induced by platelet
385 NPs in darkness (Fig. 3h).

386 **3.4 Confocal micro-Raman spectroscopy**

387 Confocal micro-Raman imaging spectroscopy was used to evaluate the uptake and the
388 distribution of the different types of TiO₂-NPs within the cells. 3D chemical images are built
389 by superimposing the different maps of each cell at their corresponding focal planes and they
390 are presented using a color meshwork i.e. blue for cell tissues and red for TiO₂ agglomerates.
391 At least five cells were analyzed to provide statistically significant results. As the sections of
392 Fig. 4 show, the uptake of the TiO₂-NPs by the cells was mainly demonstrated for P25, food
393 grade and platelet NPs (Fig. 4a,b,c) while no TiO₂ signal was registered inside the cells for
394 bypyramids and rods (Fig. 4d,e).

396 **4. Discussion**

397 The aim of this study was to investigate the cytotoxicity and genotoxicity of three different
398 shapes of TiO₂-NPs and to compare them with two commercially available TiO₂-NPs. The

399 factors taken into account for this study were: i) the physico-chemical properties of the
400 particles such as shape, particle size, agglomeration state in culture media, crystalline phase,
401 ii) the ability of the particles to induce cytotoxicity and genotoxicity, iii) the increase of the
402 toxicological effects under light exposure due to the photocatalytic activity of TiO₂ and iv)
403 the uptake of the NPs by human cells.

404 Published results on toxicity of TiO₂-NPs show high variability. Reasons for this variability
405 include physico-chemical characteristics of NPs, different methods for prepare NPs
406 dispersions, differences in NPs size and dispersion stability, and different exposure protocols
407 (Charles *et al.*, 2018). The characteristics of NPs dispersion can be influenced by medium
408 components, such as serum proteins, and by NPs properties (size, shape, surface charge,
409 surface coating etc.) (Huk *et al.* 2015). According to the study of Prasad *et al.* (2013), the
410 present results showed that in all the TiO₂-NPs dispersions, the agglomeration state increases
411 in base RPMI (without serum), while the size distribution in DMSO and in complete RPMI
412 medium (with serum) is quite similar. The different agglomeration state is probably due to
413 the ability of metal oxide NPs to adsorb proteins onto their surface, forming a “protein
414 corona” which favors less agglomeration in complete medium, which contains more proteins
415 (Prasad *et al.* 2013). Considering the results obtained, complete medium was selected as
416 cytotoxicity/genotoxicity assay medium.

417 The viability of BEAS-2B treated with commercial and engineered TiO₂-NPs after exposure
418 with light or in darkness was assessed using the WST-1 assay.

419 Commercial TiO₂-NPs induced low or no viability reduction (P25 and food grade,
420 respectively) detected by WST-1 assay, such that these results are in agreement with some
421 reports on the cytotoxicity of commercial TiO₂-NPs on BEAS-2B cells (Bhattacharya *et al.*
422 2009, Falck *et al.* 2009). Other studies showed that commercial TiO₂-NPs induced
423 cytotoxicity on BEAS-2B (Shi *et al.* 2010, Ursini *et al.* 2014). According to the results on

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424 P25, in the study of Prasad *et al.* (2013) 100 µg/ml of commercial P25 NPs produced a
425 viability decrease in BEAS-2B cells after 24h exposure. In contrast, Park *et al.* (2008) found
426 that exposure of BEAS-2B cells to commercial P25 (5-40 µg/ml) for 24h led to significant
427 cell death, both in a time- and concentration-dependent manner. Instead, fewer studies have
428 been performed using commercial food grade TiO₂-NPs. Proquin *et al.* (2017) tested these
429 NPs on different cell lines: on Caco-2, they observed cytotoxicity, while, according to our
430 results, on HCT116 they did not observe any cytotoxic effect up to the concentration of 100
431 µg/cm², confirming that the cytotoxicity can be influenced by the use of different cell lines.
432 Recently, the scientific community have produced reference NPs, which have been well
433 characterized. About evaluation of these reference TiO₂-NPs, Di Bucchianico *et al.* (2016)
434 assessed cytotoxic effects of some of these NPs (anatase 50-150 nm, anatase 5-8 nm, rutile
435 20-28 nm) in BEAS-2B cells; according to the present results the study shows in general no
436 or low effects at the doses tested (2-100 µg/ml).
437 The data demonstrated that cytotoxicity was slightly affected by light exposure, which
438 induced an increase of cellular damage after incubation with commercial P25 and engineered
439 rods.
440 Comparing the results of WST-1 assay and LDH release, the first showed low cytotoxic
441 effect at the doses tested, while the second did not show any cytotoxicity in both exposure
442 protocols. The discrepancy between LDH release and WST-1 data suggests that the viability
443 reduction may be caused by apoptosis, a cell death pathway in which the plasma membrane is
444 maintained, as observed in other studies (Schilirò *et al.* 2015). This is in accordance with
445 previous studies, which demonstrated that TiO₂-NPs could cause apoptosis in BEAS-2B cells
446 (Park *et al.* 2008, Shi *et al.* 2010).

1 447 Results of Comet assay in presence of light and in darkness showed a significant DNA
2 448 damage induced by commercial P25 and food grade NPs and engineered platelet NPs, while
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4 449 no genotoxicity was observed with other NPs.
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7 450 Considering that the uptake of NPs could involve interactions of NPs with DNA, the
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9 451 observed genotoxic effect could be related to the presence of P25, food grade and platelet
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11 452 NPs into the BEAS-2B detected in the present study and by other studies (Bhattacharya *et al.*
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13 453 2009, Park *et al.* 2008). The higher uptake of P25, food grade and platelet NPs could be
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15 454 explained with the higher agglomeration tendency (higher hydrodynamic diameter) (table 1).
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17 455 Indeed, as supposed by Magdolenova *et al.* (2012b), TiO₂-NPs that form large agglomerates,
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19 456 differently from NPs that form smaller ones, precipitate at the bottom of the cell culture
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21 457 wells, increasing the real amount of NPs to which the cells are exposed. After penetration
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23 458 into the cells, these NPs may have direct access to DNA via transport into the nucleus and/or
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25 459 during mitosis when the dissolution of nuclear membrane occurs, so they could cause DNA
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27 460 breakage (Magdolenova *et al.* 2014). Other hypothesized mechanisms that can induce the
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29 461 observed DNA damage are that NPs (or metal ions released from particles) can enhance the
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31 462 permeability of the lysosomal membrane, inducing the release of DNases (Karlsson *et al.*
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33 463 2010) or that NPs aggregates can deform nucleus causing DNA damage (Di Virgilio *et al.*
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35 464 2010).
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44 465 In addition, to quantify effects due to the photocatalytic activity of TiO₂, the highest doses
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46 466 (80, 120, 160 µg/ml) of NPs that showed a genotoxic effect were tested also in darkness
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48 467 (24h). Results showed that light exposure induced additional indirect genotoxicity,
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50 468 demonstrating a higher oxidative potential of TiO₂-NPs after exposure with light than in
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52 469 darkness. The presence of light increased DNA oxidative damage probably due to the
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54 470 photocatalytic activity of TiO₂-NPs, which caused an increase of NPs ability to produce
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56 471 radicals. In particular, based on previous studies, the anatase crystal structure of TiO₂ (used in
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472 this study) seems to be the most catalytic/photocatalytic crystalline structure of TiO₂ and
473 seems to be activated under both ultraviolet and visible light (Warheit and Donner 2015). A
474 recent study (De Matteis *et al.* 2016) demonstrated that, in particular using anatase, light is a
475 dominant factor to induce oxidative stress, TiO₂-NPs degradation and toxic effects.
476 According to the present results, Gerloff *et al.* (2009) showed direct and oxidative genotoxic
477 effects induced by TiO₂-NPs (80%/20% anatase-rutile) only in the presence of interior light.
478 Karlsson *et al.* (2008) found that TiO₂-NPs (mixture of rutile and anatase) in darkness did not
479 show oxidative DNA damage using the Fpg-Comet assay (Karlsson *et al.* 2010). However, as
480 observed in the present study, also in darkness TiO₂-NPs can induce oxidative DNA damage,
481 although lower than damage induced by light exposure. This result was observed also in the
482 study of Gurr *et al.* (2005).
483 The results obtained highlight that food grade and engineered platelet NPs induced direct
484 genotoxicity also in darkness. For the commercial food grade NPs the damage was lower than
485 in presence of light. This result agree with the study of Gopalan *et al.* (2009); they suggest
486 that TiO₂ (anatase 40 – 70 nm range) is capable of inducing higher direct genotoxic effects
487 after simultaneous irradiation with UV, respect to genotoxicity induced in darkness. The
488 increase of direct DNA damage after exposure with light, attested by Gopalan *et al.* (2009)
489 and detected for food grade NPs, remain to be explained. A possible mechanisms, that may
490 lead to this effect, could be related to the potential interaction of TiO₂-NPs with proteins
491 involved in DNA repair, as demonstrated by Jugan *et al.* (2011). Genotoxicity is not only
492 linked to the level of DNA damage but also to the type of lesions generated and their capacity
493 to be repaired. NPs exposure in presence of light could influence activity of proteins such as
494 repair enzymes, resulting in DNA damage not repaired or misrepaired (Magdolenova *et al.*
495 2014).

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496 In conclusion, cytotoxicity of NPs was low, and was influenced by the NP shape as well as
497 by light exposure. Instead, genotoxicity seemed to be influenced by the cellular-uptake and
498 the aggregation tendency of TiO₂-NPs. These two aspects are probably related to different
499 physico-chemical characteristics of NPs, such as the shape. Moreover, the presence of light
500 enhanced the genotoxic effect of some NPs primarily increasing the oxidative stress. In
501 summary, the results obtained indicate that the TiO₂-NPs shape may play a critical role in the
502 potential genotoxicity and light can influence cytotoxicity and genotoxicity of both
503 commercial and engineered TiO₂-NPs. The results obtained suggests that these shape
504 engineered TiO₂-NPs are probably safer than the commercial ones.

505 506 **Funding**

507 This work was supported by the SETNanoMetro Seventh Framework Programme project
508 (project number 604577; call identifier FP7-NMP-2013_LARGE-7).

509 510 **Competing interests**

511 The authors declare that they have no competing interests.

512 513 **References**

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706 **Table**

Sample	Particle size (nm)	D _h DMSO (nm)	D _h RPMI Base (nm)	D _h RPMI Complete (nm)	Crystalline Phase
P25	20 ± 5 quasi-spherical	107 ± 31	722 ± 246	121 ± 37	Anatase:Rutile (5:1)
Food grade	150 ± 50 undefined shape	184 ± 61	278 ± 54	184 ± 55	Anatase
Bipyramids	50 ± 9* (aspect ratio 3:2)	66 ± 20	259 ± 46	88 ± 24	Anatase

Rods	$108 \pm 47^*$	36 ± 12	1500 ± 471	39 ± 17	Anatase
	(aspect ratio 1:5)				
Platelets	$75 \pm 25^*$	233 ± 70	281 ± 83	250 ± 82	Anatase
	(aspect ratio 8:1)				

707

708 Table 1. Physico-chemical properties of the TiO₂-NPs samples. Data are presented as mean ±
709 standard deviation of 500 NPs for the particle size and 5 measurements for the hydrodynamic
710 diameter (D_h) of each sample. *The particle size was calculated along the major axis of the
711 NPs.

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720 **Figure captions**

721 Figure 1. SEM In-lens micrographs: (a) P25, (b) food grade, (c) bipyramids, (e) platelet NPs.
722 T-SEM micrograph of rods (d). DLS analyses, normalized by volume distribution (f-j): (f)
723 P25, (g) food grade, (h) bipyramids, (i) rods and (j) platelet NPs, suspensions in DMSO 1%
724 (black line), RPMI base (red line) and RPMI complete (blue line).

725 Figure 2. Cytotoxicity (a,b) and lactate dehydrogenase (LDH) release (expressed as a
726 percentage) (c,d) of BEAS-2B cells exposed to different concentrations (5–80 µg/ml) of
727 commercial and engineered NPs. Control level is at 100%. Data represent effects detected

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728 after exposure with light (a,c) and in darkness (b,d). Bars represent the mean %, error bars
729 represent standard error of mean. * $p < 0.05$; *vs* control cells (C-) according to ANOVA,
730 followed by Dunnett's test.

731 Figure 3. Effect of BEAS-2B cells exposure to commercial and engineered NPs evaluated by
732 the Comet assay (\pm Fpg). Exposure with light (a-e): (a) P25, (b) food grade, (c) bipyramids,
733 (d) rods, (e) platelet NPs; exposure in darkness (f-h): (f) P25, (g) food grade, (h) platelet NPs.
734 Bars represent the mean % of tail intensity value, error bars represent standard error of mean.
735 * $p < 0.05$ *vs* control cells (C-). # $p < 0.05$ treatment -Fpg *vs* treatment +Fpg. According to
736 ANOVA, followed by Dunnett's test.

737 Figure 4. 3D confocal micro-Raman imaging of BEAS-2B cells after exposure to commercial
738 and engineered NPs. Top views (optical and 3D Raman) and 3D Raman sections are shown
739 from the left to the right: (a) P25, (b) food grade, (c) platelet NPs, (d) bipyramids, (e) rods.
740 3D chemical images are built by superimposing the different maps of each cell at their
741 corresponding focal planes and they are presented using a color meshwork i.e. blue for cell
742 tissues (methylene blue $\nu(\text{C-C})$ ring at 1600 cm^{-1}) and red for TiO_2 agglomerates (Eg band at
743 144 cm^{-1} of the anatase TiO_2).

Figure 1

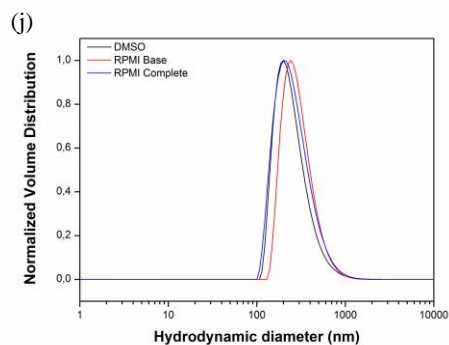
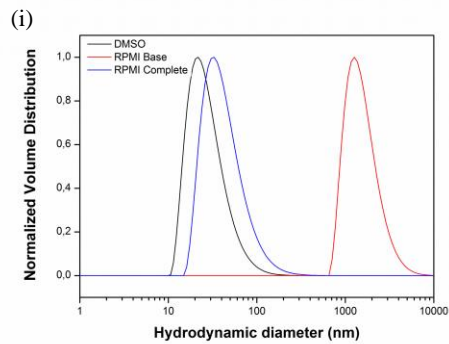
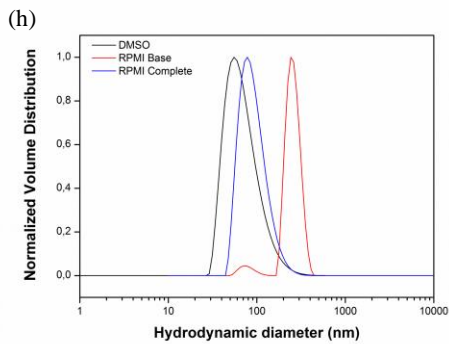
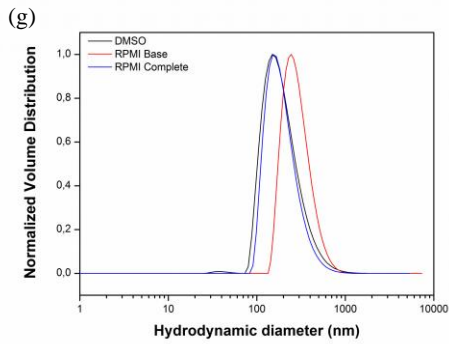
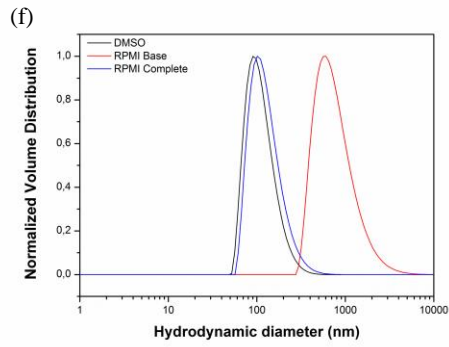
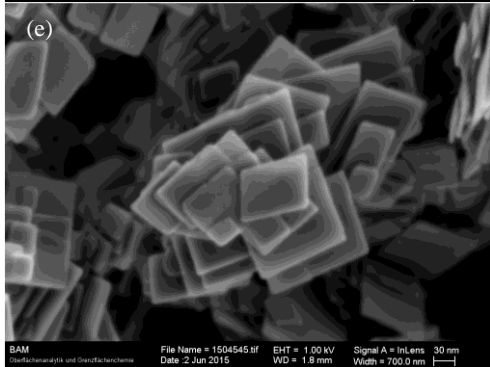
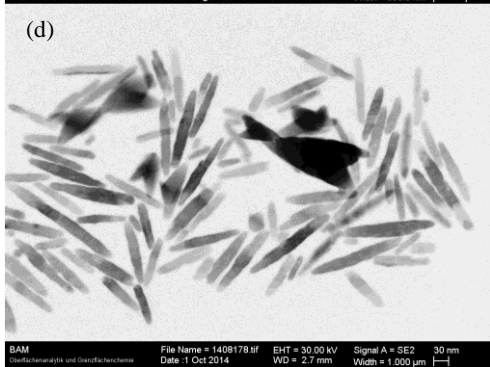
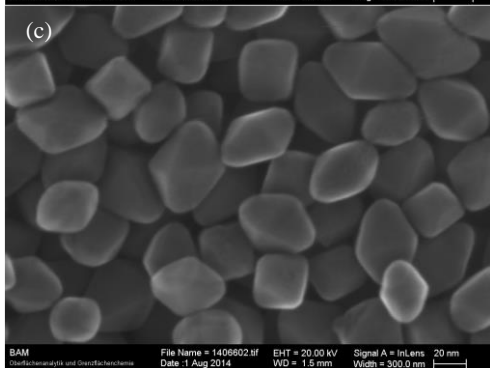
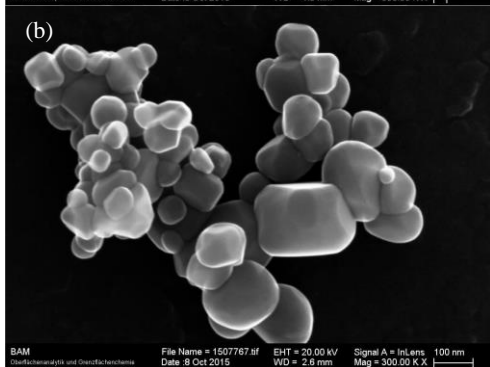
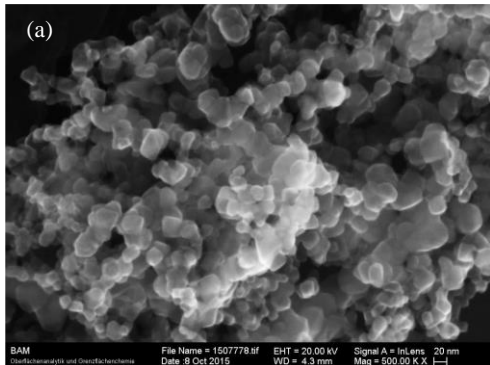
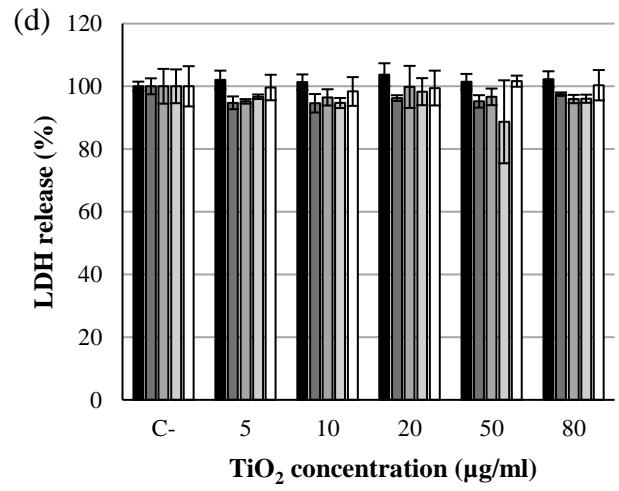
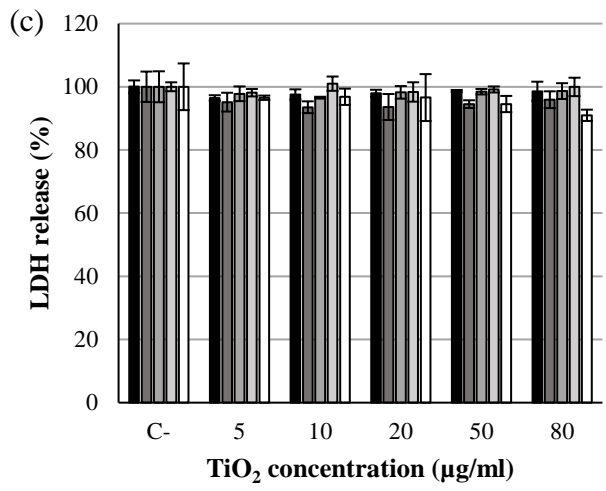
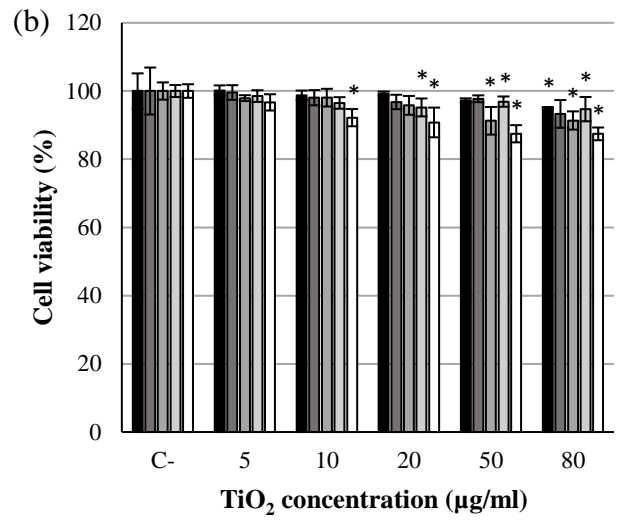
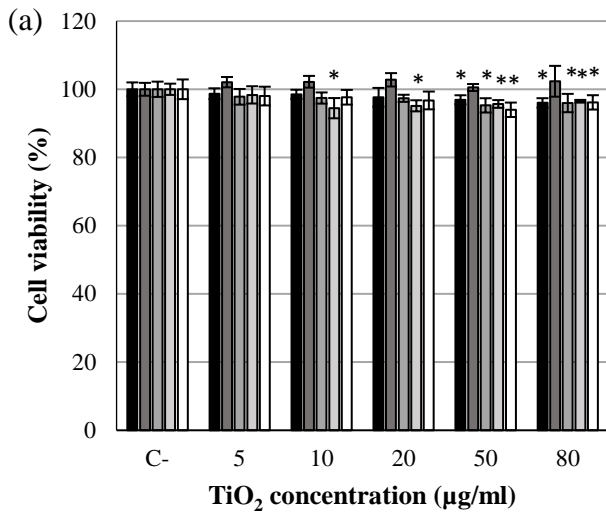


Figure 2



■ P25 ■ Food grade ■ Bipyramids □ Rods □ Platelet NPs

Figure 3

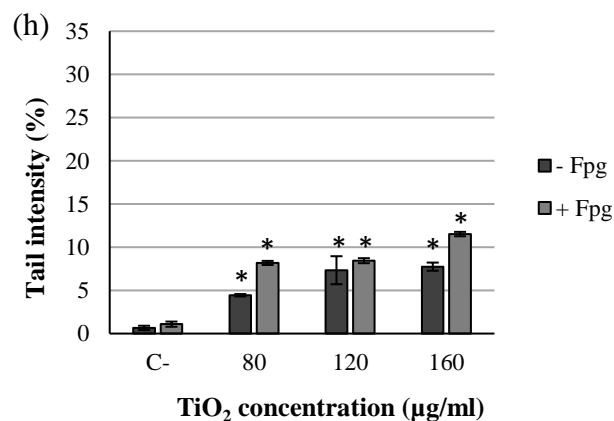
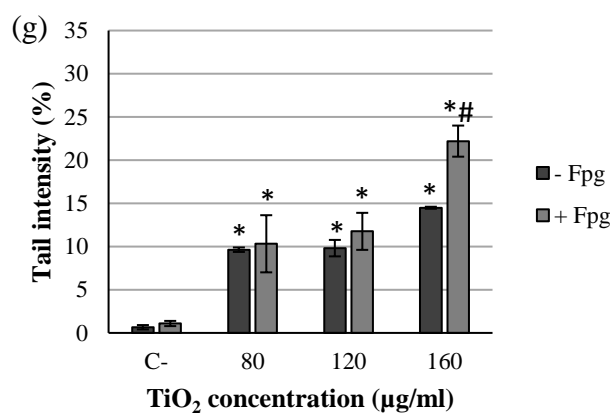
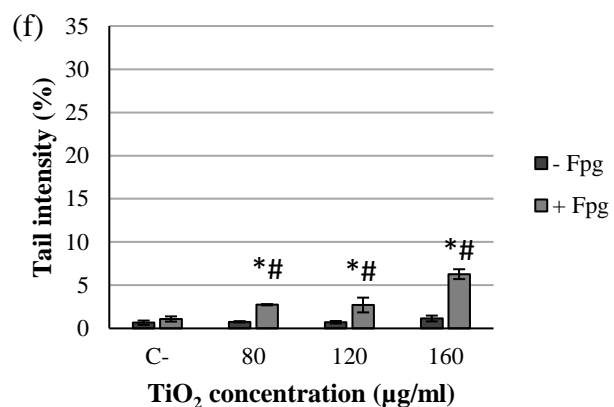
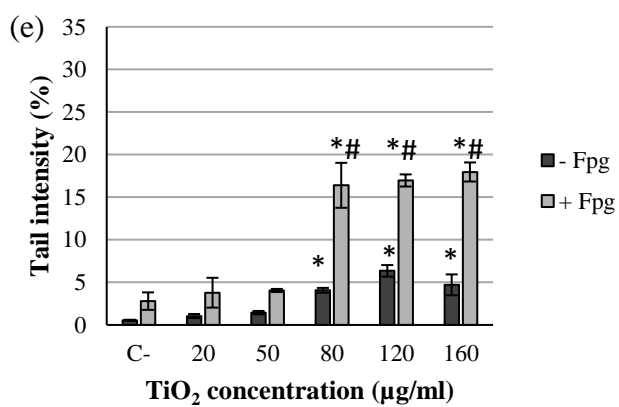
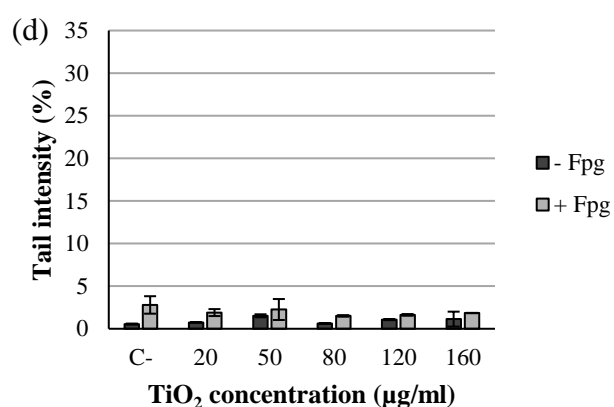
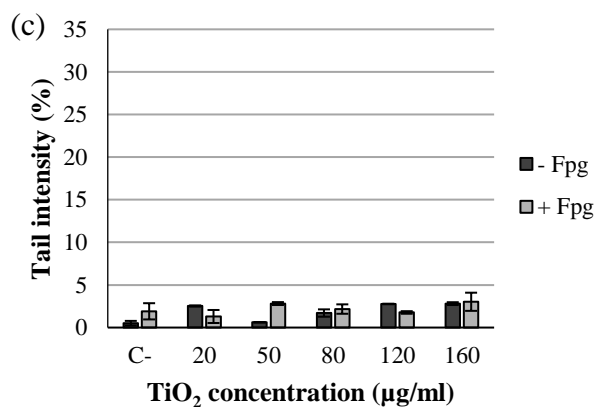
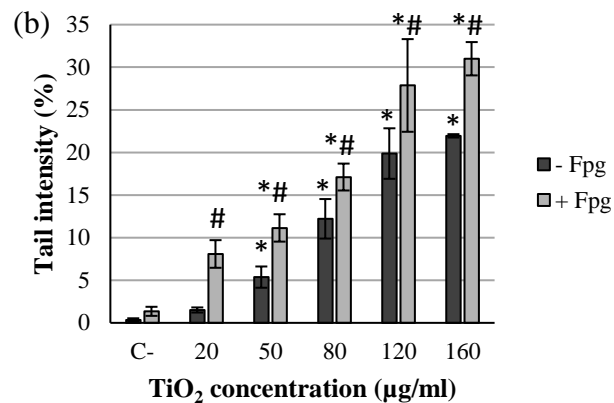
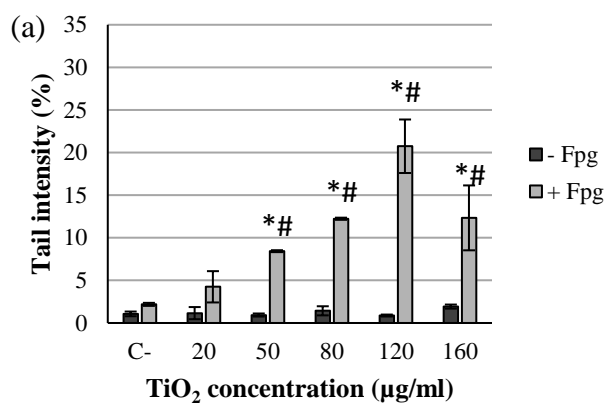
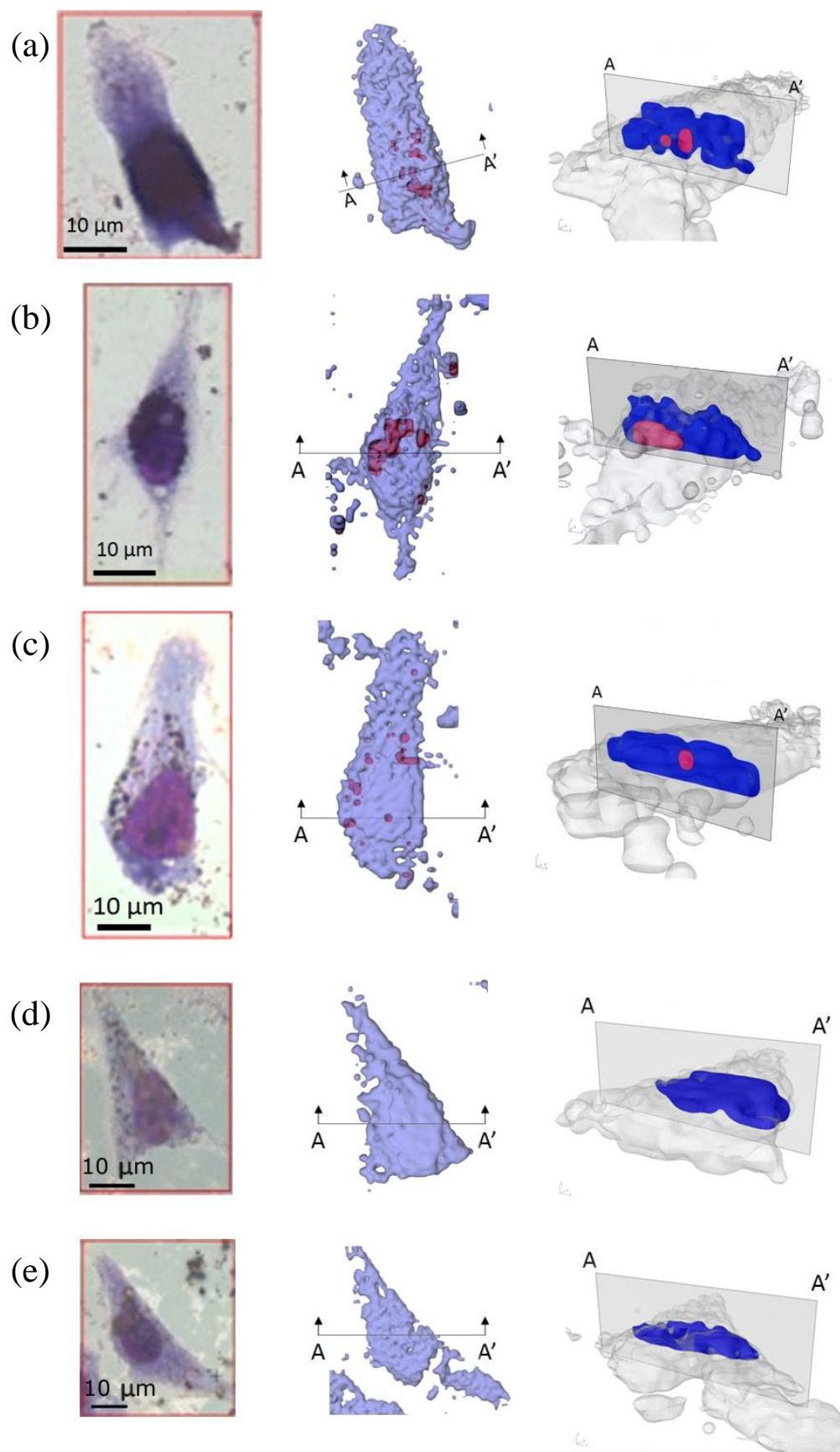


Figure 4



Supplemental

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